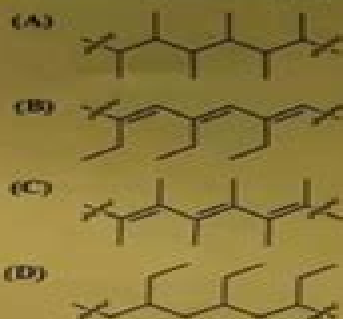


25. Which two compounds ionize with loss of a bromide ion to form the same carbocation?



- (A) I and IV
(B) II and III
(C) III and IV
(D) I and II

26. What is the product of polymerization of 1-butene?



27. Which synthetic sequence will give the highest yield of this product?



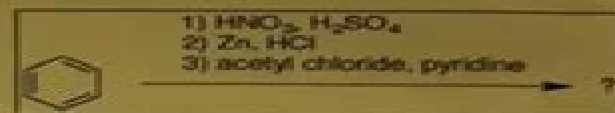
- (A) $\text{H}_3\text{C}-\text{C}\equiv\text{C}-\text{H} \xrightarrow{\text{NaNH}_2} \text{CH}_3-\text{C}(\text{Br})=\text{CH}_2 \xrightarrow{\text{H}_2, \text{Lindlar}}$
(B) $\text{H}_3\text{C}-\text{C}\equiv\text{C}-\text{H} \xrightarrow{\text{NaNH}_2} \text{CH}_3-\text{C}(\text{Br})=\text{CH}_2 \xrightarrow{\text{Na}, \text{NH}_3}$
(C) $\text{CH}_3-\text{C}\equiv\text{C}-\text{H} \xrightarrow{\text{NaNH}_2} \text{CH}_3-\text{C}(\text{Br})=\text{CH}_2 \xrightarrow{\text{H}_2, \text{Lindlar}}$
(D) $\text{CH}_3-\text{C}\equiv\text{C}-\text{H} \xrightarrow{\text{NaNH}_2} \text{CH}_3-\text{C}(\text{Br})=\text{CH}_2 \xrightarrow{\text{Na}, \text{NH}_3}$

28. What is the major product?



- (A) $\text{H}_3\text{CH}_2\text{CH}_2\text{C}(\text{OH})(\text{CH}_2\text{CH}_3)_2$
(B) $\text{H}_3\text{CH}_2\text{CH}_2\text{C}(\text{OH})(\text{CH}_2\text{CH}_3)_2$
(C) $\text{H}_3\text{CH}_2\text{CH}_2\text{C}(\text{OH})(\text{CH}_2\text{CH}_3)_2$
(D) $\text{H}_3\text{CH}_2\text{CH}_2\text{C}(\text{OH})(\text{CH}_2\text{CH}_3)_2$

29. What is the major product?



- (A) $\text{C}_6\text{H}_5\text{NHCOCH}_3$
(B) $\text{C}_6\text{H}_4(\text{NH}_2)\text{COCH}_3$
(C) $\text{C}_6\text{H}_4(\text{NH}_2)\text{CH}_2\text{CH}_3$
(D) $\text{C}_6\text{H}_4(\text{NH}_2)\text{Cl}$

30. Rank from slowest to fastest rate of electrophilic aromatic substitution.



- (A) I < II < III
(B) III < II < I
(C) II < III < I
(D) II < I < III

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A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the University of California, Santa Cruz text.

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
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










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