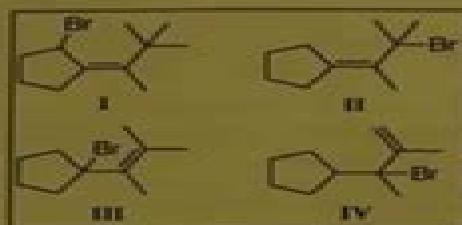


25. Which two compounds ionize with loss of a bromide ion to form the same carbocation?



(A) I and IV  
(B) II and III  
(C) III and IV  
(D) I and II

26. What is the product of polymerization of 1-butene?

(A)   
(B)   
(C)   
(D) 

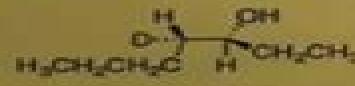
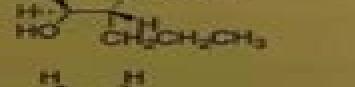
27. Which synthetic sequence will give the highest yield of this product?



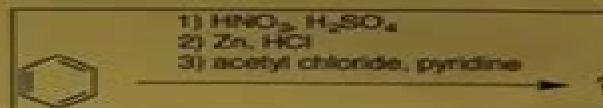
(A)  $\text{H}_3\text{C}-\text{CH}=\text{CH}_2 \xrightarrow{\text{NaNH}_2} \text{CH}_3\text{CH}(\text{Br})\text{CH}_2\text{CH}_3 \xrightarrow{\text{H}_2, \text{ Lindlar}} \text{H}_3\text{C}-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$   
(B)  $\text{H}_3\text{C}-\text{CH}=\text{CH}_2 \xrightarrow{\text{NaNH}_2} \text{CH}_3\text{CH}(\text{Br})\text{CH}_2\text{CH}_3 \xrightarrow{\text{Na, NH}_3} \text{H}_3\text{C}-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$   
(C)  $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3 \xrightarrow{\text{NaNH}_2} \text{CH}_3\text{CH}(\text{Br})\text{CH}_2\text{CH}_3 \xrightarrow{\text{H}_2, \text{ Lindlar}} \text{H}_3\text{C}-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$   
(D)  $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3 \xrightarrow{\text{NaNH}_2} \text{CH}_3\text{CH}(\text{Br})\text{CH}_2\text{CH}_3 \xrightarrow{\text{Na, NH}_3} \text{H}_3\text{C}-\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$

28. What is the major product?



(A)   
(B)   
(C)   
(D) 

29. What is the major product?



(A)   
(B)   
(C)   
(D) 

30. Rank from slowest to fastest rate of electrophilic aromatic substitution.



(A) I < II < III  
(B) III < II < I  
(C) II < III < I  
(D) I < II < III

# Acs Exam Study Guide General Organic Biochemistry

**KJ Lindholm-Leary**

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## **Table of Contents Acs Exam Study Guide General Organic Biochemistry**

1. Understanding the eBook Acs Exam Study Guide General Organic Biochemistry
  - The Rise of Digital Reading Acs Exam Study Guide General Organic Biochemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Acs Exam Study Guide General Organic Biochemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acs Exam Study Guide General Organic Biochemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acs Exam Study Guide General Organic Biochemistry
  - Personalized Recommendations
  - Acs Exam Study Guide General Organic Biochemistry User Reviews and Ratings
  - Acs Exam Study Guide General Organic Biochemistry and Bestseller Lists
5. Accessing Acs Exam Study Guide General Organic Biochemistry Free and Paid eBooks
  - Acs Exam Study Guide General Organic Biochemistry Public Domain eBooks
  - Acs Exam Study Guide General Organic Biochemistry eBook Subscription Services
  - Acs Exam Study Guide General Organic Biochemistry Budget-Friendly Options

6. Navigating Acs Exam Study Guide General Organic Biochemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Acs Exam Study Guide General Organic Biochemistry Compatibility with Devices
  - Acs Exam Study Guide General Organic Biochemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Acs Exam Study Guide General Organic Biochemistry
  - Highlighting and Note-Taking Acs Exam Study Guide General Organic Biochemistry
  - Interactive Elements Acs Exam Study Guide General Organic Biochemistry
8. Staying Engaged with Acs Exam Study Guide General Organic Biochemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Acs Exam Study Guide General Organic Biochemistry
9. Balancing eBooks and Physical Books Acs Exam Study Guide General Organic Biochemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Acs Exam Study Guide General Organic Biochemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Acs Exam Study Guide General Organic Biochemistry
  - Setting Reading Goals Acs Exam Study Guide General Organic Biochemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acs Exam Study Guide General Organic Biochemistry
  - Fact-Checking eBook Content of Acs Exam Study Guide General Organic Biochemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
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