

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (iii) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acids Bases And Solutions Answer Key

Darrell D. Ebbing



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Table of Contents Acids Bases And Solutions Answer Key

1. Understanding the eBook Acids Bases And Solutions Answer Key
 - The Rise of Digital Reading Acids Bases And Solutions Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acids Bases And Solutions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acids Bases And Solutions Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acids Bases And Solutions Answer Key
 - Personalized Recommendations
 - Acids Bases And Solutions Answer Key User Reviews and Ratings
 - Acids Bases And Solutions Answer Key and Bestseller Lists
5. Accessing Acids Bases And Solutions Answer Key Free and Paid eBooks
 - Acids Bases And Solutions Answer Key Public Domain eBooks
 - Acids Bases And Solutions Answer Key eBook Subscription Services
 - Acids Bases And Solutions Answer Key Budget-Friendly Options
6. Navigating Acids Bases And Solutions Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Acids Bases And Solutions Answer Key Compatibility with Devices
 - Acids Bases And Solutions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acids Bases And Solutions Answer Key
 - Highlighting and Note-Taking Acids Bases And Solutions Answer Key
 - Interactive Elements Acids Bases And Solutions Answer Key
8. Staying Engaged with Acids Bases And Solutions Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Acids Bases And Solutions Answer Key
- 9. Balancing eBooks and Physical Books Acids Bases And Solutions Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acids Bases And Solutions Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Acids Bases And Solutions Answer Key
 - Setting Reading Goals Acids Bases And Solutions Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acids Bases And Solutions Answer Key
 - Fact-Checking eBook Content of Acids Bases And Solutions Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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