

Appendix 1

Abbreviations and symbols for quantities and units

Unavoidably, some symbols are used for more than one thing. The meaning is usually clear from the context, but always check that you understand what a symbol is representing. Note that symbols representing *physical quantities* are printed in *italics*. Symbols for units are *not* in italics. The reference (in brackets) given for each entry is to chapters where the abbreviation or symbol is used.

A	ampere (unit of electrical current) (1, 14)	°C	degrees Celsius (1)
A	absorbance (10, 11, 28)	C_p	specific heat capacity (13)
A	area (8, 13, Masses, Isotopes, MTS)	c	concentration of a solution (1)
A	frequency factor (in Arrhenius equation) (9)	c	speed of light (2, 19)
A	Madelung constant (6)	c	root mean square speed (of molecules) (20)
A_r	relative atomic mass (1)	c	mean speed (of molecules) (20)
Å	angstrom (non-SI unit used for bond length) (1)	CPSE	crystal field stabilization energy (29)
a	acceleration (1, 8)	C <small>l</small>	chemical ionization (in mass spectrometry) (32)
a	thermodynamic activity (14, 15, 16, 37)	CEP	Cahn–Ingold–Prelog sequence rules for R/S nomenclature of chiral centres (18)
a	van der Waals constant for gases (8)	C <small>br</small>	carboxybenzyl group (24)
a_0	Bohr radius (30)	cm	centimetre (unit of length) (1)
AAS	atomic absorption spectrometry (11)	cm ³	cubic centimetre (unit of volume) (1)
AES	atomic emission spectrometry (11)	cm ⁻¹	reciprocal centimetre (unit of wavenumber) (10, 12)
AO	atomic orbital (4)	D	debye (non-SI unit of electric dipole moment) (8, 17)
Ac	acetyl (2)	D	bond dissociation enthalpy (4, 13)
Ar	general aryl group (2)	D	mean bond enthalpy (13)
[acac] ⁺	pentane-2,4-dicarboxy-(acetylacetoneato) (28)	D	centrifugal distortion constant (in rotational spectroscopy) (10)
atm	atmosphere (non-SI unit of pressure) (1, 8)	d	distance (1)
a	magnetic field strength (30)	D <small>b</small>	doublet (in NMR spectrum) (12)
B	rotational constant (10)	DBU	1,8-diazabicyclo[5.4.0]undec-7-ene (19)
b	van der Waals constant for gases (8)	DMAP	4-dimethylaminopyridine (19)
Bn	benzyl (2)	DMP	<i>N,N</i> -dimethylformamide (2, 20)
Boc	tert-butyloxycarbonyl group (24)	DMSO	dimethyl sulfoxide (20)
Bz	benzoyl (2)	D <small>a</small>	dalton (atomic mass unit) (1)
bar	bar (unit of pressure) (1, 8)	d <small>en</small>	1,4,7-azatetralin (diethylenetriamine) (28)
bipy	2,2'-bipyridine (28)	dm	cubic decimetre (unit of volume) (1)
bp	bonding pair (5)	E	electrophile (19)
b.p.	normal boiling point (usually quoted as °C) (1)	E	energy (3, 10)
br	broad (IR absorption) (12)	E_{ex}	kinetic energy (1, 10)
C	coulomb (unit of electric charge) (1, 16)	E_{in}	potential energy (1, 10, 17)
C	Curie constant (in Curie's law for magnetic susceptibility) (28)		
C_p	molar heat capacity at constant pressure (13, 14)		
C_V	molar heat capacity at constant volume (13, 14)		

Appendix D Chemistry Answers

A Gutmann

Appendix D Chemistry Answers:

Chemistry John Olmsted, Greg Williams, Robert C. Burk, 2020 Chemistry 4th Edition is an introductory general chemistry text designed specifically with Canadian professors and students in mind. A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers and distinguish this text from other offerings. It more accurately reflects the curriculum of most Canadian institutions. Chemistry is sufficiently rigorous while engaging and retaining student interest through its accessible language and clear problem solving program without an excess of material and redundancy.

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Chemistry James N. Spencer, George M. Bodner, Lyman H. Rickard, 1999 CD ROM contains study tool illustrations, photos, definitions, text, interactive notebook drawing tools.

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General Organic and Biological Chemistry Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing, dental hygiene, nutrition, medical technology and occupational therapy. It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry. Students need have no previous background in chemistry but should possess basic math skills. The text features numerous helpful problems and learning features.

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