

Name _____ Date _____ Per _____

2.2 Section Review

1. What is the difference between a heterogeneous and a homogeneous mixture?
2. Describe a procedure that could be used to separate a mixture consisting of sand and salt.
3. Classify each of the following as a substance or a mixture:
 - a.) silver
 - b.) alphabet soup
 - c.) textbook
 - d.) table salt (sodium chloride)
4. Describe in your own words the difference between a pure substance and a mixture.
5. Describe ways in which the various components of a mixture can be separated.
6. Explain the term phase as it relates to homogeneous and heterogeneous mixtures.

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R Sanford



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