

## PHYSICAL SCIENCE PRACTICE TEST WITH ANSWER KEY

1. A weak attraction between hydrogen atom in one molecule and an oxygen in other called \_\_\_\_\_.  
A. covalent bond  
B. Hydrogen bond  
C. Ionic bond  
D. Metallic bond
2. Which of the following substances is most likely a covalent compound?  
A.  $\text{CaCO}_3$   
B.  $\text{KNO}_3$   
C.  $\text{SO}_2$   
D.  $\text{NaOH}$
3. What is the element needed to bond with Na to create the most ionic bond?  
A. F  
B. Cl  
C. Br  
D. I
4. The boiling point of substance A is higher than that of substance B. What does this mean?  
A. Substance A is less viscous than substance B.  
B. Substance A evaporates faster than substance B.  
C. Substance B has greater surface tension than substance A.  
D. Substance B has weaker intermolecular forces than substance A.
5. What is the intermolecular force of attraction between the nucleus of one atom and the electrons of a nearby atom?  
A. Dipole-dipole force  
B. Ion-dipole force  
C. London dispersion force  
D. Van der Waals force
6. Which of the following has the greatest number of atoms?  
A.  $\text{Ca(OH)}_2$   
B.  $\text{Al}_2\text{O}_3$

# Apex Answer Key Physical Science Semester 1

**SA Dillow**



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