

Name: \_\_\_\_\_

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# Acids and Bases Strength

Write True (T) or False (F).

- I. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. \_\_\_\_\_
- II. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. \_\_\_\_\_

Write equations that show how the following reactions occur in water.

- I.  $\text{F}^-$  acts as a base. \_\_\_\_\_
- II.  $\text{HNO}_2$  acts as an acid. \_\_\_\_\_
- III.  $\text{Fe}(\text{H}_2\text{O})_5^{3+}$  acts as an acid. \_\_\_\_\_
- IV.  $\text{HCO}_3^-$  acts as a base. \_\_\_\_\_
- V.  $\text{HCO}_3^-$  acts as an acid. \_\_\_\_\_
- VI.  $\text{Al}(\text{H}_2\text{O})_5(\text{OH})^{2+}$  acts as a base. \_\_\_\_\_

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

- I. 0.001 M HCl
- II. 0.76 M KOH
- III.  $2.8 \times 10^{-4}$  M  $\text{Ba}(\text{OH})_2$

# Acid Base Worksheets With Answers

**Elizabeth L. Hammerman**

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