

Name : \_\_\_\_\_ Date : \_\_\_\_\_

# Acids and Bases Strength

Write True (T) or False (F).

- i. Stronger acids have weaker conjugate bases and weaker acids have stronger conjugate bases. \_\_\_\_\_
- ii. Stronger bases have stronger conjugate acids and weaker bases have weaker conjugate acids. \_\_\_\_\_

Write equations that show how the following reactions occur in water.

- i.  $F^-$  acts as a base. \_\_\_\_\_
- ii.  $HNO_2$  acts as an acid. \_\_\_\_\_
- iii.  $Fe(H_2O)_6^{3+}$  acts as an acid. \_\_\_\_\_
- iv.  $HCO_3^-$  acts as a base. \_\_\_\_\_
- v.  $HCO_3^-$  acts as an acid. \_\_\_\_\_
- vi.  $Al(H_2O)_5(OH)^{2+}$  acts as a base. \_\_\_\_\_

What volume of 0.45 M LiOH would be needed to neutralize 60.0 mL of 0.15 M HI?

Calculate the pH values of the following solutions of strong acids and bases.

- i. 0.001 M HCl
- ii. 0.76 M KOH
- iii.  $2.8 \times 10^{-4}$  M  $Ba(OH)_2$

# Acid Base Worksheets With Answers

**Elizabeth L. Hammerman**



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