

Acid-Base balance Questions and answers

Respiratory acidosis **CORRECT ANSWERS** pH low, PaCO₂ high, K⁺ high,
compensated by HCO₃ high to bring pH to normal

Metabolic acidosis **CORRECT ANSWERS** pH low, HCO₃ low, K⁺ high
compensated by PaCO₂ low

Respiratory alkalosis **CORRECT ANSWERS** pH high, PaCO₂ low, K⁺ low,
compensated by HCO₃ low to bring pH to normal

Metabolic alkalosis **CORRECT ANSWERS** pH high, HCO₃ high, K⁺ low
compensated by PaCO₂ high

Buffer system **CORRECT ANSWERS** $\text{CO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_2\text{CO}_3 \rightleftharpoons \text{H}^+ + \text{HCO}_3^-$

What causes

Metabolic alkalosis? **CORRECT ANSWERS** loss of hydrogen ions or acids or
an excess of base bicarbonate

For example:

- hypovolemia,
- the loss of gastric fluid via NG suction or vomiting,
- excessive bicarbonate intake,
- massive transfusion of whole blood,
- hyperaldosteronism.

The nurse reviews the arterial blood gas results of a client and notes the following: pH 7.45, PaCO₂ of 30 mm Hg (30 mm Hg), and HCO₃⁻ of 20 mEq/L (20 mmol/L). The nurse analyzes these results as indicating which condition? **CORRECT ANSWERS** Respiratory alkalosis, compensated

Acid Base Question Packet

L Cohen



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