

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (iii) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Acid And Base Answer Key

Carolina Biological Supply Company

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the Carolina Biological Supply Company text.

Acid And Base Answer Key:

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Table of Contents Acid And Base Answer Key

1. Understanding the eBook Acid And Base Answer Key
 - The Rise of Digital Reading Acid And Base Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acid And Base Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acid And Base Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acid And Base Answer Key
 - Personalized Recommendations
 - Acid And Base Answer Key User Reviews and Ratings
 - Acid And Base Answer Key and Bestseller Lists
5. Accessing Acid And Base Answer Key Free and Paid eBooks

- Acid And Base Answer Key Public Domain eBooks
- Acid And Base Answer Key eBook Subscription Services
- Acid And Base Answer Key Budget-Friendly Options
- 6. Navigating Acid And Base Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Acid And Base Answer Key Compatibility with Devices
 - Acid And Base Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acid And Base Answer Key
 - Highlighting and Note-Taking Acid And Base Answer Key
 - Interactive Elements Acid And Base Answer Key
- 8. Staying Engaged with Acid And Base Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acid And Base Answer Key
- 9. Balancing eBooks and Physical Books Acid And Base Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acid And Base Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Acid And Base Answer Key
 - Setting Reading Goals Acid And Base Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Acid And Base Answer Key
 - Fact-Checking eBook Content of Acid And Base Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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