

# Unit 1



# UNIT 1: UNDERSTANDING PERIODICITY

## Section 1.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|       |   |
|-------|---|
| _____ | 1 |
| _____ | 2 |
| _____ | 3 |
| _____ | 4 |
| _____ | 5 |

|                        |       |
|------------------------|-------|
| Atomic Mass            | _____ |
| Atomic Number          | _____ |
| Electron Configuration | _____ |
| Chemical Name          | _____ |
| Chemical Symbol        | _____ |

Put each item in Column A, either the letter of the matching item in Column B.

### Column A

- \_\_\_\_\_ 16. It exists in the periodic table.
- \_\_\_\_\_ 17. It has no line periodic table.
- \_\_\_\_\_ 18. Group 1 elements.
- \_\_\_\_\_ 19. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 20. Group 2 elements.

### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 21. There are two representative elements of elements.
- \_\_\_\_\_ 22. Most elements have lines in the periodic table.
- \_\_\_\_\_ 23. Group 1 elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 24. Group 2 elements are the alkaline earth metals.
- \_\_\_\_\_ 25. Group 1 elements are highly reactive elements in the periodic table.
- \_\_\_\_\_ 26. Group 2 elements are very reactive elements found in the periodic table.
- \_\_\_\_\_ 27. Elements have properties of both metals and non-metals.

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