

(B) 68

(C) 34

(D) 33

Q.No. 95 An unbiased coin is tossed. If the outcome is a head then a pair of unbiased dice is rolled and the sum of the numbers obtained on them is noted. If the toss of the coin results in tail then a card from a well-shuffled pack of nine cards numbered 1, 2, 3, ..., 9 is randomly picked and the number on the card is noted. The probability that the noted number is either 7 or 8 is :

(A) 19/72

(B) 19/36

(C) 15/72

(D) 13/36

Q.No. 96 A metal ball of mass 0.1 kg is heated upto 500°C and dropped into a vessel of heat capacity 800J K^{-1} and containing 0.5 kg water. The initial temperature of water and vessel is 30°C . What is the approximate percentage increment in the temperature of the water ? [Specific Heat Capacities of water and metal are, respectively, $4200\text{J kg}^{-1}\text{K}^{-1}$ and $400\text{J kg}^{-1}\text{K}^{-1}$]

(A) 30%

(B) 20 %

(C) 25 %

(D) 15%

Q.No. 97 If $\text{Re}\left(\frac{z-1}{2z+i}\right) = 1$ where $z = x+iy$, then the point (x,y) lies on a

(A) circle whose centre is $(-3/2, -3/2)$

(B) straight line whose slope is $-(3/2)$

(C) straight line whose slope is $3/2$

(D) circle whose diameter is $\sqrt{5}/2$



I



II



III

Q.No. 98 Arrange the following amines in the decreasing order of basicity :

(A) I > III > II

(B) III > II > I

(C) III > I > II

(D) I > II > III

Q.No. 99 The equation of the line passing through $(-4, 3, 1)$, parallel to the plane $x+2y-z-5=0$ and intersecting the line $\frac{x+1}{-3} = \frac{y-2}{2} = \frac{z-1}{-1}$ is

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 net ionic equation $\text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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 $\text{AgNO}_3 \text{aq} + \text{KCl} \text{aq} \rightarrow \text{AgCl} \text{s} + \text{KNO}_3 \text{aq}$ total ionic $\text{Ag}^+ \text{aq} + \text{NO}_3^- \text{aq} + \text{K}^+ \text{aq} + \text{Cl}^- \text{aq} \rightarrow \text{AgCl} \text{s} + \text{K}^+ \text{aq} + \text{NO}_3^- \text{aq}$ net ionic $\text{Ag}^+ \text{aq} + \text{Cl}^- \text{aq} \rightarrow \text{AgCl} \text{s}$ 2

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