

8th Grade Science Chapter 12 Answers

1. The periodic table was created by a Russian chemist, Dmitri Mendeleev in the 1860's. P 302
2. The term periodic means, "to have a regular or repeating pattern."
3. The elements are arranged in a Periodic Table because the properties of these elements change in a periodic (regular or repeating pattern) way. Think about the calendar, every week has 7 days, it is a "period" of time.
4. Mendeleev arranged his elements in order of their increasing atomic mass. P 303
5. Mosley helped to correct Mendeleev's order by atomic mass. Instead of using atomic mass, Mosley determined that it was best to use the elements atomic number or the number of protons in the atom. Once Mosley was done, all of the elements fell into their proper place on the table.
6. The Periodic Law is a principle that states that the chemical and physical properties of the elements are periodic functions of their atomic numbers.
7. the 6 = atomic #
 C = symbol
 Carbon = Element Name
 12.0 = atomic mass
8. The background colors of the Periodic Table actually indicate what type of element it is. The Blue background is for a metals, the Green's are metalloids, and the yellow indicates a nonmetal.
9. The color of the chemical symbol indicates the physical state of the element at room temperature. If the symbol is: Red = Solid Blue = Liquid Green = Gas
10. A row of elements is called a Period.
11. A column of elements is called a Group or Family
12. The number in the parenthesis is the mass number of the most stable isotope of the element.
13. Yes, the elements of the symbols are universally recognized by an International Committee of scientists that approves the name and symbol for each element.
14. Elements are classified as a metal, nonmetal, or a metalloid, according to their properties.
15. The number of electrons in the outer energy level is used to determine which category an element is placed.
16. Four properties of metals are: Good Conductors of Thermal Energy, Malleable: they can be flattened with a hammer without shattering, Ductile: they can be drawn into thin wires and are then good conductors of electric current, Shiny: you can see a reflection.
17. Nonmetals are: Poor Conductors of Thermal Energy: can't conduct thermal energy, Not Malleable: can't be flattened with a hammer without shattering, Not Ductile: can not be drawn into wires, & Not Shiny.

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