

# Activity Series

To become familiar with the relative activities of metals in chemical reactions.

## Apparatus

small test tubes\* (13)

test tube rack

## Chemicals

0.2 M  $\text{Ca}(\text{NO}_3)_2$

0.2 M  $\text{Zn}(\text{NO}_3)_2$

0.2 M  $\text{FeSO}_4$

0.2 M  $\text{CuSO}_4$

7 small pieces each of calcium, magnesium, zinc, iron wool, tin, and copper

0.2 M  $\text{Mg}(\text{NO}_3)_2$

0.2 M  $\text{Fe}(\text{NO}_3)_3$

0.2 M  $\text{SnCl}_4$

6 HCl

## OBJECTIVE

## APPARATUS

## AND CHEMICALS

Chemical elements are equally classified by their properties into three groups: metals, nonmetals, and metalloids. Most of the known elements are metals. Their physical properties include high thermal and electrical conductivity, high luster, malleability (ability to be pounded flat without shattering), and ductility (ability to be drawn out into a fine wire). All common metals are solids at room temperature except mercury, which is a liquid. The periodic table illustrated in Figure 15.1 shows the classification of the elements.

All elements to the left of the shaded area except hydrogen are metals. Those to the right are nonmetals. Those in the shaded area have intermediate properties and are called semimetals or metalloids. Families or groups of elements consist of elements in vertical columns in the periodic table. Elements within a group or family (called congeners) have similar chemical properties because they have similar valence electronic structures; that is, the number of valence electrons (electrons in the outermost shell) is the same for all members of a family or group. For historical reasons, most of the groups have names, some often referred to by their names. These are the following:

1. Group 1A, called *alkali metals* because they react with oxygen to form bases
2. Group 2A, called *alkaline earth metals* because their presence makes soils
3. Group 3A, no common name
4. Group 4A, no common name

## DISCUSSION

\*A spot plate may be used in place of test tubes.

# Activity Series Post Lab Answers Experiment 7

**Zhe Xu**



## **Activity Series Post Lab Answers Experiment 7:**

**Resources for Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents

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