

MELT

15 Based on Table S, an atom of which element has the strongest attraction for electrons in a chemical bond?

- (1) chlorine
- (2) nitrogen

- (3) oxygen
- (4) selenium

EN

16 At which temperature and pressure would a sample of helium behave most like an ideal gas?

- (1) 100. K and 500. kPa
- (2) 50. K and 500. kPa
- (3) 300. K and 50. kPa
- (4) 600. K and 50. kPa

high T  
Low P

systems tend to undergo changes toward

- (1) lower entropy and lower energy
- (2) lower entropy and high energy
- (3) higher entropy and lower energy
- (4) higher entropy and high energy

no  
crazy  
nocturnal

21 Amines, amides, and amino acids are categories of

- (1) isomers
- (2) isotopes
- (3) organic compounds
- (4) inorganic compounds

22 A molecule of which compound has a multiple covalent bond?

- (1)  $\text{CH}_4$
- (2)  $\text{C}_2\text{H}_4$
- (3)  $\text{C}_3\text{H}_8$
- (4)  $\text{C}_4\text{H}_{10}$

# August 13 2009 Chemistry Regents Answers

**A Loxley**



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