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A group of graduates in blue gowns and caps are celebrating on a stage. Some are holding diplomas, and one student in the center is throwing their cap into the air. The background is dark, and the stage floor is reflective.

ACADEMIC SUCCESS MEDIA

# Act Compass Study Guide

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in the n 2 molecule the nitrogen atoms have an  $\sigma$  bond and two  $\pi$  bonds holding the two atoms together the presence of three strong bonds makes n 2 a very stable molecule phosphorus is a third period element and as such does not form  $\pi$  bonds efficiently therefore it must fulfill its bonding requirement by forming three  $\sigma$  bonds

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complete atoms can or electrons to get a stable outer energy level gain lose share a figure that uses the symbol of the element and dots to represent the electrons in the outer energy level also used to show how valence electrons are bonded when elements combine to form compounds

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