

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Block: \_\_\_\_\_

## Practice Acid/Base Quiz

**Directions:** Answer all questions on this practice quiz. Use this as a resource to study for the assessment on Friday, February 28<sup>th</sup> in addition to your lecture note guides and unit study guide. Your assessment will include both multiple choice and short answer questions.

1. Which of the following could have a pH of 10?

- a. HCl                                      b.  $\text{HCH}_2\text{COO}$                                       c.  $\text{NH}_4\text{OH}$                                       d. HF

2. Which of the following is a neutralization reaction?

- a.  $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$                                       c.  $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   
b.  $\text{Ca}(\text{OH})_2 + \text{HNO}_3 \rightarrow \text{H}_2\text{O} + \text{Ca}(\text{NO}_3)_2$                                       d.  $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$

3. Which of the following is directly responsible for acid rain?

- a. steam vented from a nuclear power plant  
b. sulfur dioxide released from a coal-fired power plant  
c. mining of coal for a coal-fired power plant  
d. processing of uranium for a nuclear power plant

Use the following reaction to answer questions 4 through 8.



4. Which compound represents the acid?

- a. LiOH                                      b.  $\text{H}_2\text{O}$                                       c. LiF                                      d. HF

5. Which compound represents the salt?

- a. LiOH                                      b.  $\text{H}_2\text{O}$                                       c. LiF                                      d. HF

6. Which compound represents the base?

- a. LiOH                                      b.  $\text{H}_2\text{O}$                                       c. LiF                                      d. HF

7. Which compound has the highest number of hydrogen ( $\text{H}^+$ ) ions?

- a. LiOH                                      b.  $\text{H}_2\text{O}$                                       c. LiF                                      d. HF

8. Which compound has the highest number of hydroxide ( $\text{OH}^-$ ) ions?

- a. LiOH                                      b.  $\text{H}_2\text{O}$                                       c. LiF                                      d. HF

# Acid Base Reactions Multiple Choice Questions

**Brian Murphy, Clair Murphy, Brian J.  
Hathaway**



## Acid Base Reactions Multiple Choice Questions:

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