

## Chemistry 113 Practice Exam 4 Spring 2016

Directions on actual Exam: The exam consists of 2 sections (Chemical Bonding and Geometry and IMFs. Each multiple choice question is worth 4.33 points. You are allowed to use a scientific (nonprogrammable) calculator, periodic table, and VSEPR Theory Table.

Useful information:

$$\Delta H^\circ_{rxn} = \Sigma n \times D(\text{bonds broken}) - \Sigma n \times D(\text{bonds formed})$$

where  $\Sigma n$  represents the sum of the moles of a particular type of bond

$D$  represents bond energy per mole of bond

Chemical Bonding and Geometry

1.

The Lewis dot symbol for the calcium ion is

A)

$2+$

B)

Ca

C)

$2+$

D)

Ca $2+$

E)

Ca

## 9 Spring Practice Exam

**Kathleen Armour**

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