

CHAPTER 4 REVIEW***Arrangement of Electrons in Atoms*****SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. In what way does the photoelectric effect support the particle theory of light?

2. What is the difference between the ground state and the excited state of an atom?

3. Under what circumstances can an atom emit a photon?

4. How can the energy levels of the atom be determined by measuring the light emitted from an atom?

5. Why does electromagnetic radiation in the ultraviolet region represent a larger energy transition than does radiation in the infrared region?

5 2 Section Review Electrons Arrangement In Atoms

Robert R. Redfield



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