

General Instructions :

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises 15 questions. All questions are compulsory.
- (ii) This question paper is divided into three sections – **A**, **B** and **C**.
- (iii) **Section A** – Questions No. 1 to 7 are short answer type questions. Each question carries 2 marks.
- (iv) **Section B** – Questions No. 8 to 13 are also short answer type questions. Each question carries 3 marks.
- (v) **Section C** – Questions No. 14 and 15 are case-based questions. Each question carries 4 marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

1. (a) Write the molecular formula of the following carbon compounds :
 (i) Methane
 (ii) Propane
 (b) Carbon compounds have low melting and boiling points. Why ? 2
2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 (a) Write the group numbers of these elements in the Modern Periodic Table.
 (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 (c) Which of the two is electropositive ? 2
3. (a) Which of the following flowers will have higher possibility of self-pollination ?
 Mustard, Papaya, Watermelon, Hibiscus
 (b) List the two reproductive parts of a bisexual flower. 2
4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2

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WJ Hussar

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2physical Science Grade Question Paper Book Review: Unveiling the Magic of Language

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Table of Contents 2physical Science Grade Question Paper

1. Understanding the eBook 2physical Science Grade Question Paper
 - The Rise of Digital Reading 2physical Science Grade Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 2physical Science Grade Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2physical Science Grade Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2physical Science Grade Question Paper
 - Personalized Recommendations
 - 2physical Science Grade Question Paper User Reviews and Ratings
 - 2physical Science Grade Question Paper and Bestseller Lists

5. Accessing 2physical Science Grade Question Paper Free and Paid eBooks
 - 2physical Science Grade Question Paper Public Domain eBooks
 - 2physical Science Grade Question Paper eBook Subscription Services
 - 2physical Science Grade Question Paper Budget-Friendly Options
6. Navigating 2physical Science Grade Question Paper eBook Formats
 - ePUB, PDF, MOBI, and More
 - 2physical Science Grade Question Paper Compatibility with Devices
 - 2physical Science Grade Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2physical Science Grade Question Paper
 - Highlighting and Note-Taking 2physical Science Grade Question Paper
 - Interactive Elements 2physical Science Grade Question Paper
8. Staying Engaged with 2physical Science Grade Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2physical Science Grade Question Paper
9. Balancing eBooks and Physical Books 2physical Science Grade Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2physical Science Grade Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 2physical Science Grade Question Paper
 - Setting Reading Goals 2physical Science Grade Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 2physical Science Grade Question Paper
 - Fact-Checking eBook Content of 2physical Science Grade Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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