

What's Inside?

Preview of 1 of 8 Topics

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Topic
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WATER PROPERTIES & ITS ROLE ON EARTH

HOW DOES WATER'S UNIQUE BEHAVIOR SHAPE EARTH AND SUPPORT LIFE?

What You Need to Know

- Water is the **universal solvent** for most chemical reactions.
- The **hydrogen bonds** that naturally occur are **weak bonds** compared to covalent bonds.
- These molecules often **cooperatively hydrogen bond** with each other.
- Surface tension** makes water form droplets and menisci.
- Water has a **high heat capacity** and **high heat of fusion** and **high heat of vaporization**.
- Thermal energy** causes water to change its state, affecting the way it behaves.
- Water can absorb a tremendous amount of thermal energy.
- Large bodies of water release tremendous heat energy slowly.
- Water is essential for the **survival** of life.
- Physical** properties include boiling point, freezing point, etc.
- Chemical** properties include reactivity, etc.
- Water is essential for **forming** clouds and precipitation.
- Water is essential for **transporting** nutrients.

Real-World Science Spotlight

Why is it that water has a high heat capacity?

Water has a high heat capacity because of the hydrogen bonds between water molecules. These bonds are weak, but they are numerous, and they require a lot of energy to break. This means that water can absorb a lot of heat without its temperature rising significantly.

Vocabulary to Know

- Solvent** - a substance that dissolves another substance.
- Solute** - a substance that is dissolved in a solvent.
- Surface Tension** - the cohesive forces between liquid molecules.
- Thermal Energy** - the energy that causes particles to move.
- Phase Change** - the change of state from one to another.
- Thermal Expansion** - the increase in volume of a substance as its temperature rises.
- Physical Property** - a characteristic that can be observed without changing the substance.
- Chemical Property** - a characteristic that can only be observed by changing the substance.
- Hydrogen Bonds** - weak bonds between water molecules.
- Evaporation** - the process of a liquid turning into a gas.

Models & Visuals to Know

Cohesion, Adhesion, and Surface Tension



WATER PROPERTIES & ITS ROLE

How does water's unique behavior shape Earth and support life?

Water is the universal solvent for most chemical reactions. The hydrogen bonds that naturally occur are weak bonds compared to covalent bonds. These molecules often cooperatively hydrogen bond with each other. Surface tension makes water form droplets and menisci. Water has a high heat capacity and high heat of fusion and high heat of vaporization. Thermal energy causes water to change its state, affecting the way it behaves. Water can absorb a tremendous amount of thermal energy. Large bodies of water release tremendous heat energy slowly. Water is essential for the survival of life. Physical properties include boiling point, freezing point, etc. Chemical properties include reactivity, etc. Water is essential for forming clouds and precipitation. Water is essential for transporting nutrients.

Why is it that water has a high heat capacity? Water has a high heat capacity because of the hydrogen bonds between water molecules. These bonds are weak, but they are numerous, and they require a lot of energy to break. This means that water can absorb a lot of heat without its temperature rising significantly.

What is the difference between cohesion and adhesion? Cohesion is the force that holds water molecules together, while adhesion is the force that holds water molecules to a surface.

What is surface tension? Surface tension is the force that holds water molecules together at the surface of the liquid.

What is thermal energy? Thermal energy is the energy that causes particles to move.

What is a phase change? A phase change is the change of state from one to another.

What is thermal expansion? Thermal expansion is the increase in volume of a substance as its temperature rises.

What is a physical property? A physical property is a characteristic that can be observed without changing the substance.

What is a chemical property? A chemical property is a characteristic that can only be observed by changing the substance.

What are hydrogen bonds? Hydrogen bonds are weak bonds between water molecules.

What is evaporation? Evaporation is the process of a liquid turning into a gas.

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Unit 10 -
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