



THE PROPERTIES OF MATTER

Review Worksheet

1. What are the properties of matter? Properties of matter are the characteristics that distinguish one substance from another.

- a. What is mass? Mass is the amount of matter in an object. What is volume? Volume is the amount of space an object takes up. What is density? Density is the mass of an object divided by its volume. What is temperature? Temperature is a measure of the average kinetic energy of the particles in a substance. What is the SI unit for mass? Kilogram (kg). What is the SI unit for volume? Cubic meter (m³). What is the SI unit for density? Kilogram per cubic meter (kg/m³). What is the SI unit for temperature? Celsius (°C).
- b. What are the three states of matter? Solid, liquid, and gas. What are the properties of solids? Solids have a fixed shape and volume. The particles in a solid are packed closely together and vibrate in place. What are the properties of liquids? Liquids have a fixed volume but no fixed shape. The particles in a liquid are close together but can move past each other. What are the properties of gases? Gases have no fixed shape or volume. The particles in a gas are far apart and move rapidly in all directions.
- c. What is a mixture? A mixture is a combination of two or more substances that are not chemically bonded together. What is a pure substance? A pure substance is a material that has a constant composition and distinct chemical and physical properties. What is an element? An element is a substance that cannot be broken down into simpler substances by chemical means. What is a compound? A compound is a substance formed by the chemical combination of two or more elements in a fixed ratio.
- d. What is a physical property? A physical property is a characteristic that can be observed or measured without changing the identity of the substance. What is a chemical property? A chemical property is a characteristic that can only be observed when a substance undergoes a chemical change. What is a physical change? A physical change is a change in the form or appearance of a substance without changing its chemical identity. What is a chemical change? A chemical change is a change in the chemical identity of a substance.
- e. What is a mixture? A mixture is a combination of two or more substances that are not chemically bonded together. What is a pure substance? A pure substance is a material that has a constant composition and distinct chemical and physical properties. What is an element? An element is a substance that cannot be broken down into simpler substances by chemical means. What is a compound? A compound is a substance formed by the chemical combination of two or more elements in a fixed ratio.

21 Properties Of Matter Answers

Cathy Humphreys, Ruth E. Parker

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