

Determining Salt Concentration Using Two Methods

Name _____

Instructor _____ Section _____ Partner _____

(20 pts for the first table; 12 pts for the second table)

Data**Density**

Tables of mass and density values for the salt solutions.

Salt Solution Percentage (use the values on the stock bottles)	Mass of the empty beaker	Mass of 10.00mL solution with the beaker	Mass of 10.00mL of the solution alone	Density of the solution in g/mL
2.000 %	77.161 g	87.527 g		
4.000 %	77.481 g	87.614 g		
6.000 %	77.523 g	87.832 g		
8.000 %	77.506 g	87.710 g		
10.000 %	77.312 g	87.722 g		

Unknown Solution	Mass of the empty beaker	Mass of 10.00mL solution with the beaker	Mass of 10.00mL of the solution alone	Density of the unknown solution in g/mL
A	77.623 g	87.633 g		
B	77.451 g	87.595 g		
C	77.311 g	87.304 g		

Answers Key To Chemistry 1211 Lab Manual

**Oak Ridge National Laboratory.
Analytical Chemistry Division**



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