



Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Limiting Reactants

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

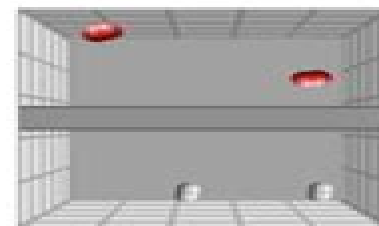
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make? \_\_\_\_\_ you can make 30 hotdogs with single bun and wieners \_\_\_\_\_
2. Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make \_\_\_\_\_
3. Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more. \_\_\_\_\_

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$ . The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of  **$\text{O}_2$  molecules** and  **$\text{H}_2$  molecules** to two.
  - A. How many hydrogen molecules ( $\text{H}_2$ ) are there? \_\_\_\_\_ 1 \_\_\_\_\_ Hydrogen atoms? \_\_\_\_\_ 2 \_\_\_\_\_
  - B. How many oxygen molecules ( $\text{O}_2$ ) are there? \_\_\_\_\_ 1 \_\_\_\_\_ Oxygen atoms? \_\_\_\_\_ 2 \_\_\_\_\_
2. How many  $\text{H}_2\text{O}$  molecules do you think will form when these four molecules react? \_\_\_\_\_ 4 \_\_\_\_\_



# Answers For Student Limiting Reactants

**J Elliott**



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