



Student Exploration: Limiting Reactants

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

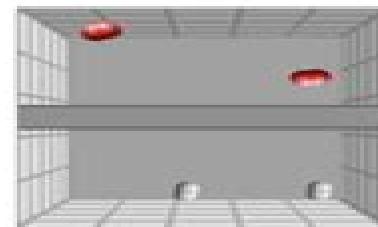
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make? _____ you can make 30 hotdogs with single bun and wieners _____
2. Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make _____
3. Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more. _____

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure $H_2 + O_2$ becomes H_2O . The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of O_2 molecules and H_2 molecules to two.
 - A. How many hydrogen molecules (H_2) are there? _____ 1 _____ Hydrogen atoms? _____ 2 _____
 - B. How many oxygen molecules (O_2) are there? _____ 1 _____ Oxygen atoms? _____ 2 _____
2. How many H_2O molecules do you think will form when these four molecules react?
_____ 4 _____



Answers For Student Limiting Reactants

J Elliott

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