

Acids and Bases

Section Quiz: Acid-Base Theories

In the space provided, indicate which of the theories predicts that each compound meets the criteria for an acidic or basic compound.

1. HCl (strong acid) - Lowery theory.
a. not applicable (indicated)
b. both an acid and a base (indicated)
c. **acidic potential**
d. HCl is the answer
2. H_2SO_4 is not the following, but it is a **strong acid**. Lowery theory would not indicate a base? **Indicate your answer by checking the correct box.**
a. H_2SO_4
b. **HSO_4^-**
c. H_2SO_3
d. HSO_3^-
3. Is the following reaction, which is indicated, correct as a **strong acid**? **Indicate your answer.**
a. **HCl**
b. H_2SO_4
c. H_2O
d. H_2O_2
4. Is the following reaction, which is indicated, correct as a **strong acid**? **Indicate your answer.**
a. HCl
b. **H_2SO_4**
c. H_2O
d. H_2O_2
5. H_2SO_4 is not the following, but it is a **polyprotic acid**.
a. HCl
b. **H_2SO_4**
c. H_2SO_3
d. H_2O

Acids And Bases Section Quiz Propertiex Of Acids

Rolland J. Gladieux



Acids And Bases Section Quiz Properties Of Acids:

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