

# Electron Configuration (Ions / Valence) - Worksheet #6

Symbol	# of Electrons	Electron Configuration (Use EITHER method, Core or "Long")
$Ca^{+2}$	18	$Ca^{+2} : [Ar]^{18}$
$Se^{-2}$	32	$Se^{-2} : [Ar]^{18} 3s^2 3p^6$
$Sn^{+2}$	22	$Sn^{+2} : [Kr]^{18} 5s^2 4s^2$
$Zn^{+2}$	10	$Zn^{+2} : [Ar]^{10} 3d^10$
$Sr^{+2}$		$Sr^{+2} : [Kr]^{18}$
$O^{-2}$	6	$O^{-2} : [He]^{2} 2s^2 2p^4$
$Cs^{+1}$		$Cs^{+1} : [Xe]^{18}$
$I^{-}$	8	$I^{-} : [Kr]^{18} 5s^2 4s^2 3p^6$
$Fe^{+3}$	7	$Fe^{+3} : [Ar]^{14} 4s^1 3d^5$
$Cu^{+2}$	9	$Cu^{+2} : [Ar]^{10} 4s^2 3d^9$

Directions: State how many valence electrons

1. O \_\_\_\_\_ 6 \_\_\_\_\_

2. P \_\_\_\_\_ 5 \_\_\_\_\_

3. V \_\_\_\_\_

4. Ca \_\_\_\_\_ 2 \_\_\_\_\_

5. Xe \_\_\_\_\_ 0 \_\_\_\_\_

6. Hg \_\_\_\_\_

7. Te \_\_\_\_\_ 6 \_\_\_\_\_

8. Cl<sup>-</sup> \_\_\_\_\_

9. I<sup>-3</sup> \_\_\_\_\_

10. Zn<sup>+2</sup> \_\_\_\_\_

11. Ge<sup>-4</sup> \_\_\_\_\_

12. Sb<sup>+3</sup> \_\_\_\_\_

13. O<sup>-</sup> \_\_\_\_\_

14. Tc<sup>-4</sup> \_\_\_\_\_

## 5 3 Electron Configurations Answer Key

**Disha Experts**

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### **Table of Contents 5 3 Electron Configurations Answer Key**

1. Understanding the eBook 5 3 Electron Configurations Answer Key
  - The Rise of Digital Reading 5 3 Electron Configurations Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying 5 3 Electron Configurations Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 5 3 Electron Configurations Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from 5 3 Electron Configurations Answer Key
  - Personalized Recommendations

- 5 3 Electron Configurations Answer Key User Reviews and Ratings
- 5 3 Electron Configurations Answer Key and Bestseller Lists

5. Accessing 5 3 Electron Configurations Answer Key Free and Paid eBooks

- 5 3 Electron Configurations Answer Key Public Domain eBooks
- 5 3 Electron Configurations Answer Key eBook Subscription Services
- 5 3 Electron Configurations Answer Key Budget-Friendly Options

6. Navigating 5 3 Electron Configurations Answer Key eBook Formats

- ePUB, PDF, MOBI, and More
- 5 3 Electron Configurations Answer Key Compatibility with Devices
- 5 3 Electron Configurations Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of 5 3 Electron Configurations Answer Key
- Highlighting and Note-Taking 5 3 Electron Configurations Answer Key
- Interactive Elements 5 3 Electron Configurations Answer Key

8. Staying Engaged with 5 3 Electron Configurations Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers 5 3 Electron Configurations Answer Key

9. Balancing eBooks and Physical Books 5 3 Electron Configurations Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection 5 3 Electron Configurations Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine 5 3 Electron Configurations Answer Key

- Setting Reading Goals 5 3 Electron Configurations Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of 5 3 Electron Configurations Answer Key

- Fact-Checking eBook Content of 5 3 Electron Configurations Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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