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## Section 4.2 The Structure of an Atom- Physical Science

1. What are 3 subatomic particles? - answer a. Protons  
b. Electrons  
c. Neutrons
2. Circle the letter that identifies a subatomic particle with a positive (+) charge. -  
answer b. Proton
3. Why did Chadwick conclude that the particles produced by his experiment were  
neutral in charge? - answer A charged object did not deflect their path
4. Circle the letters of properties that vary among subatomic particles. - answer a. Color  
c. Charge  
d. Location in the Atom
5. Circle the letter of the expression that accurately compares the masses of neutrons  
and protons. - answer a. Mass of 1 neutron=mass of 1 proton
6. Two atoms of the same element can have different numbers of protons. - answer True
7. What is an atomic number? - answer The number of protons in an atom
8. Circle the letters that identify quantities that are always equal to an element's atomic  
number. - answer b. Number of Protons  
c. Number of Neutrons  
d. Number of Electrons
9. Two different elements can have the same atomic number. - answer True
10. What is the mass number of an atom? - answer The sum of the protons and  
neutrons in an atom  
(# of protons + # of neutrons)
11. Complete the equation... - answer # of neutrons=mass # - atomic # (# of protons)
12. Every atom of a given element has the same number of \_\_\_\_\_ and  
\_\_\_\_\_. - answer Protons and Electrons

# Answers To Section 2 The Atom Packet

**John R. Holum**



## Answers To Section 2 The Atom Packet:

**Regents Chemistry--Physical Setting Power Pack Revised Edition** Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key *Compact First Student's Pack (Student's Book Without Answers with CD-ROM, Workbook Without Answers with Audio CD)* Peter May, 2012-09-06 A highly focused Cambridge English First FCE course providing efficient exam preparation in 50 60 core hours The syllabus for this exam has changed and this book has now been replaced by 9781107428485 Compact First Second edition Student's Pack Student's Book without answers with CD ROM Workbook without answers with Audio **CliffsNotes Chemistry Practice Pack** Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **Quantum Theory and Measurement** John Archibald Wheeler, Wojciech Hubert Zurek, 2014-07-14 The forty nine papers collected here illuminate the meaning of quantum theory as it is disclosed in the measurement process Together with an introduction and a supplemental annotated bibliography they discuss issues that make quantum theory overarching principle of twentieth century physics appear to many to prefigure a new revolution in science Originally published in 1983 The Princeton Legacy Library uses the latest print on demand technology to again make available previously out of print books from the distinguished backlist of Princeton University Press These editions preserve the original texts of these important books while presenting them in durable paperback and hardcover editions The goal of the Princeton Legacy Library is to vastly increase access to the rich scholarly heritage found in the thousands of books published by Princeton University Press since its founding in 1905 *McGraw Hill TEAS 2-Book*

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**Elements of General and Biological Chemistry** John R. Holum,1975 Concerning goals and methods The nature of matter The atomic theory The nature of matter Compounds and chemical bonds Moles states of matter chemical energy Solutions and colloids Important ionic compounds and their reactions Acidity Detection control measurement Nonmetallic elements of the biosphere Introduction to organic chemistry Hydrocarbons Organic derivatives of water hydrogen sulfide and ammonia alcohols ethers mercaptans and amines Carbonyl compounds aldehydes and ketones carboxylic acids esters and amides Carbohydrates Lipids Proteins Enzymes vitamins hormones poisons and drugs Important fluids of the body Energy for living The metabolism of carbohydrates The metabolism of lipids The metabolism of proteins The chemistry of heredity Radiations and human health Exponentials Iupac nomenclature for common oxygen derivatives of hydrocarbons      Harcourt Science: Life science, units A and B ,2002

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