


Directions: Complete the table below by writing the name of each of the substances above under the proper heading. Place substances with a pH lower than 3.0 in the strong acids column. Place substances with a pH higher than 10.0 in the strong bases column.

1. Strong acids	2. Weak acids	3. Weak bases	4. Strong bases
			

Directions: Answer the following questions on the lines provided.

5. Is pure water an acidic, basic, or neutral substance? neutral
6. Is the pH of a strong acid higher or lower than the pH of a weak acid of the same concentration?
The pH of a strong acid is lower than that of a weak acid.
7. Is the pH of a strong base higher or lower than the pH of a weak base of the same concentration?
The pH of a strong base is higher than that of a weak base.
8. On the pH scale, what are the values of acids and what are the values of bases?
The pH of acids is less than 7; the pH of a base is greater than 7.

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