

Half-Life

Textbook pages 302–311

Before You Read

Write a sentence in the lines below explaining what the word decay means to you. As you read about radioactive decay, think about how the common meaning of decay differs from the scientific meaning.

What is radiocarbon dating?

Radiocarbon dating is the process of determining the age of an object by measuring the amount of carbon-14 remaining in that object. Carbon's isotopes include carbon-12 and carbon-14. When an organism is alive, the ratio of carbon-14 atoms to carbon-12 atoms in the organism remains nearly constant. But when an organism dies, its carbon-14 atoms decay without being replaced. The ratio of carbon-14 to carbon-12 then decreases with time. By measuring this ratio, the age of an organism's remains can be estimated. Only material from plants and animals that lived within the past 50 000 years contains enough carbon-14 to be measured using radiocarbon dating. ✓

What is a half-life and how is it used in radiocarbon dating?

A half-life is a measure of the rate of radioactive decay for a given isotope. It is equal to the time required for half the nuclei in a sample to decay. Its value is a constant for any radioactive isotope. For example, the half-life of the radioisotope strontium-90 is 29 years. If you have 10.0 g of strontium-90 today, 29 years from now you will have 5.00 g left. This is because one half-life will have passed ($10.0 \text{ g} \times \frac{1}{2} = 5.00 \text{ g}$). 58 years from now, two half-lives will have passed and 2.50 g of the sample will remain ($10.0 \text{ g} \times \frac{1}{2} \times \frac{1}{2} = 2.50 \text{ g}$). The shorter the half-life is, the faster the decay rate. A **decay curve** is a curved line on a graph that shows the rate at which radioisotopes decay.



Mark the Text

In Your Own Words

After you read this section, go back and summarize the main concepts in your own words.



Reading Check

1. Which carbon isotope undergoes radioactive decay?
-

Answers To Half Life 7 2 Summary

Ralph H. Petrucci, Robert K. Wismer



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Table of Contents Answers To Half Life 7 2 Summary

1. Understanding the eBook Answers To Half Life 7 2 Summary
 - The Rise of Digital Reading Answers To Half Life 7 2 Summary
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers To Half Life 7 2 Summary
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers To Half Life 7 2 Summary
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers To Half Life 7 2 Summary
 - Personalized Recommendations
 - Answers To Half Life 7 2 Summary User Reviews and Ratings

- Answers To Half Life 7 2 Summary and Bestseller Lists
- 5. Accessing Answers To Half Life 7 2 Summary Free and Paid eBooks
 - Answers To Half Life 7 2 Summary Public Domain eBooks
 - Answers To Half Life 7 2 Summary eBook Subscription Services
 - Answers To Half Life 7 2 Summary Budget-Friendly Options
- 6. Navigating Answers To Half Life 7 2 Summary eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers To Half Life 7 2 Summary Compatibility with Devices
 - Answers To Half Life 7 2 Summary Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers To Half Life 7 2 Summary
 - Highlighting and Note-Taking Answers To Half Life 7 2 Summary
 - Interactive Elements Answers To Half Life 7 2 Summary
- 8. Staying Engaged with Answers To Half Life 7 2 Summary
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers To Half Life 7 2 Summary
- 9. Balancing eBooks and Physical Books Answers To Half Life 7 2 Summary
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers To Half Life 7 2 Summary
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers To Half Life 7 2 Summary
 - Setting Reading Goals Answers To Half Life 7 2 Summary
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers To Half Life 7 2 Summary
 - Fact-Checking eBook Content of Answers To Half Life 7 2 Summary
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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