

Student Exploration: Chemical Changes

Summary: In this activity, students explore chemical changes by observing the formation of gases, color changes, and temperature changes. They also learn about the conservation of mass and the law of conservation of energy.

Key Concepts: Chemical changes, conservation of mass, conservation of energy, chemical reactions, chemical formulas, chemical equations, chemical properties, chemical changes, chemical reactions, chemical formulas, chemical equations, chemical properties, chemical changes.



1. A chemical change is a process in which one or more substances are converted into one or more new substances. The process of a chemical change is called a chemical reaction.

2. Chemical changes are often accompanied by the formation of a gas, a color change, or a temperature change. The formation of a gas is called effervescence.

Conservation of Mass: The law of conservation of mass states that the total mass of the reactants in a chemical reaction is equal to the total mass of the products.

Conservation of Energy: The law of conservation of energy states that the total energy of the reactants in a chemical reaction is equal to the total energy of the products.

Chemical Formulas: Chemical formulas are used to represent the composition of a chemical compound. They consist of the chemical symbols of the elements in the compound, with subscripts indicating the number of atoms of each element. For example, the chemical formula for water is H_2O , which means that one molecule of water consists of two hydrogen atoms and one oxygen atom.

Chemical Equations: Chemical equations are used to represent a chemical reaction. They show the reactants on the left side of the equation and the products on the right side. The equation is balanced when the number of atoms of each element is the same on both sides.



3. The law of conservation of mass states that the total mass of the reactants in a chemical reaction is equal to the total mass of the products.

4. The law of conservation of energy states that the total energy of the reactants in a chemical reaction is equal to the total energy of the products.

Chemical Properties: Chemical properties are the characteristics of a substance that determine how it will react with other substances. They are often used to identify a substance.



Answers To Evidence For Chemical Change Lab

J. A. Beran, Mark Lassiter



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