

Chemistry Project Lab Phase-II Answers



QUESTION 1: What is the difference between a saturated and unsaturated solution? Explain with an example. How does the solubility of a substance change with increasing temperature? Why does the solubility of a substance increase with increasing temperature? Name two substances that will dissolve more in an unsaturated solution than in the solid. Indicate the solubility process. Name the various factors that affect solubility.

ANSWER: A saturated solution is a solution in which no more solute can be dissolved. An unsaturated solution is a solution in which more solute can be dissolved. For example, a saturated solution of salt in water will not dissolve any more salt, whereas an unsaturated solution of salt in water will dissolve more salt.

ANSWER: Solubility increases with increasing temperature. This is because more solute can be dissolved in a solution at higher temperatures.

Answers To Generator Phet Lab

LL Leslie

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