

Ion Practice

Note:

1. In a neutral atom, the number of Protons and Electrons are **equal** and is the same number as the atomic number for that element.
2. Protons can never be lost or gained by an atom, which is why they are used to identify the atom.
3. Electrons can be lost or gained. When this happens atoms will get a charge because protons and electrons are no longer equal. Positive ions are called **Cations**, negative ions are called **Anions**.

1. What is an ion?
2. When an atom is an ion, where is the charge written?

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained
Ex.	Fluorine	F ⁻	9	10	Gained one
1			53	54	
2			16		Gained two
3	Potassium				Lost one
4		Ca ⁺²			
5			35	36	
6		Sr ⁻²			
7		H ⁺			
8			8		Gained two
9			12		Lost two
10	Aluminum	Al ⁺³	13	10	Lost three
11			34	36	
12		H ⁻			
13	Lithium				Lost one
14		Rb ⁺			
15			17	18	

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