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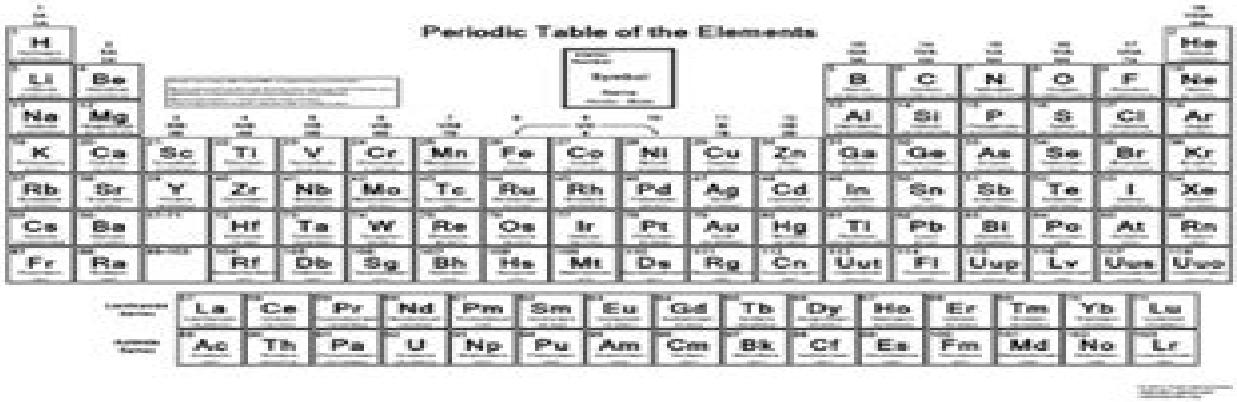


### 🔬 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🧑‍🔬✨.

### 💡 Matter:

Anything that has **mass** and **occupies space** 🏋️🌍

*Examples*: air, gases, liquids, solidz!

### ⚛️ Atoms:

Atoms are the **smallest, indivisible particle** 🔬✂️.

- **Outer part** → *shell/orbit/energy level* → carries the electronz (e−)
- **Inner part** → *nucleus* → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the **subatomic particles**

### 🧪 Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|---|---|---|---|---|
| **Proton** | $p^+$ | 1 | +1 | Nucleus |
| **Neutron** | n | 1 | 0 | Nucleus |
| **Electron** | $e^-$ | ~0 | -1 | Orbit/Shell |

# Section 2nuclear Chemistry Study Guide Key

Kenneth W. Raymond

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