

1. Fill in the chart below.

Name of element	Atomic number	# of protons	# of electrons	Category of element (metal, nonmetal, metalloid)	Valence electrons (1 to 8/valence 8) (1 to 8 larger than 8)	# of orbitals (one to four)	Bohr model of the element Draw Beryllium & Neon in the space below
Beryllium	4	4	4	Metal	2	2	
Silicon	14	14	14	Metalloid	4	3	
Neon	10	10	10	Non-metal	8	2	
Aluminum	13	13	13	Metal	3	3	
Carbon	6	6	6	Non-metal	4	2	

2. Orbits are also called energy levels

3. Most of an atom's mass is found in the Nucleus which is made up of protons & neutrons

4. ID the number of elements, molecules, & atoms in the compounds below

Compound	# of elements	Molecules	Element	atoms	Element	Atoms	Element	Atoms
NaCl	2	1	Iron	1	Oxygen	2	Hydrogen	2
Al ₂ O ₃	2	1	Aluminum	2	Sulfur	2	Oxygen	6
Mg ₃ O ₂	2	1	Magnesium	3	Chlorine	2	Oxygen	4
Li ₂ SO ₄	4	1	Sulfur	1	Phosphorus	1	Oxygen	4
Fe ₂ O ₃	3	1	Magnesium	1	Sulfur	3	Oxygen	9

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Brendan G. Carr

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