

# AP® CHEMISTRY 2019 SCORING GUIDELINES

## Question 1



Ethanol,  $\text{C}_2\text{H}_5\text{OH}$ , will combust in air according to the equation above.

(a) Is  $\text{O}_2(g)$  oxidized in the reaction, or is it reduced? Justify your answer in terms of oxidation numbers.

$\text{O}_2$  is reduced; its oxidation number changes from 0 in  $\text{O}_2$  to  $-2$  in  $\text{CO}_2$  and  $\text{H}_2\text{O}$ .

1 point is earned for the correct answer with an appropriate justification.

(b) When a sample of  $\text{C}_2\text{H}_5\text{OH}$  was combusted, the volume of  $\text{CO}_2(g)$  produced was 18.0 L when measured at 21.7°C and 1.03 atm. Determine the number of moles of  $\text{CO}_2(g)$  that was produced.

$$n = \frac{PV}{RT} = \frac{(1.03 \text{ atm})(18.0 \text{ L})}{(0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1})(273.15 + 21.7 \text{ K})} = 0.766 \text{ mol CO}_2$$

1 point is earned for the correct calculation.

(c) Determine the volume of  $\text{C}_2\text{H}_5\text{OH}(l)$ , in mL, that was combusted to produce the volume of  $\text{CO}_2(g)$  collected in part (b). (The density of  $\text{C}_2\text{H}_5\text{OH}(l)$  is 0.79 g/mL.)

$$0.766 \text{ mol CO}_2 \times \frac{1 \text{ mol C}_2\text{H}_5\text{OH}}{2 \text{ mol CO}_2} \times \frac{46.07 \text{ g C}_2\text{H}_5\text{OH}}{1 \text{ mol C}_2\text{H}_5\text{OH}} \times \frac{1 \text{ mL}}{0.79 \text{ g}} = 22 \text{ mL}$$

1 point is earned for the correct stoichiometry for moles of  $\text{C}_2\text{H}_5\text{OH}$  and  $\text{CO}_2$ .

1 point is earned for the correct volume of  $\text{C}_2\text{H}_5\text{OH}$ .

(d) Determine the amount of heat, in kJ, that is released by the combustion reaction.

$$q = n \Delta H_{\text{comb}}^\circ$$

$$q = 0.766 \text{ mol CO}_2 \times \frac{1 \text{ mol}_{\text{rxn}}}{2 \text{ mol CO}_2} \times \frac{-1270 \text{ kJ}}{1 \text{ mol}_{\text{rxn}}} = -486 \text{ kJ}$$

The amount of heat released is 486 kJ.

1 point is earned for a correct calculation (negative sign is not required).

The combustion reaction occurred in a closed room containing  $5.56 \times 10^4 \text{ g}$  of air originally at 21.7°C. Assume that all of the heat produced by the reaction was absorbed by the air (specific heat =  $1.005 \text{ J/g } ^\circ\text{C}$ ) in the room.

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**The Must-Read : Colleen Hoover's "It Ends with Us"** This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed.

**Uncover the Best : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo"** This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery.

**Discover the Magic : Delia Owens' "Where the Crawdads Sing"** This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting.

These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered.

The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. *The Secret History* is a masterful and suspenseful novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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