

Mole Ratios

How can the coefficients in a chemical equation be interpreted?

NATALIA
LACAYO
P. 2

Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10^{23} or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

Model 1 – A Chemical Reaction



1. Consider the reaction in Model 1.

a. What are the coefficients for each of the following substances in the reaction?

1 N_2

3 H_2

2 NH_3

b. Draw particle models below to illustrate the reaction in Model 1.



2. Consider each situation below as it relates to the reaction in Model 1.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole numbers possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$ (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1	3	2	1:3:2
If the reaction occurred one hundred times, how many molecules would be consumed or produced?	100	300	200	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538	1614	1076	1:3:2

Assignment 6 Mole Ratios Answer Key

Chemical Society (Great Britain)



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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs **Mo Molybdenum** Hermann Jehn, Wolfgang Kurtz, Dietrich Schneider, Ursula Trobisch, Joachim Wagner, 2013-11-11 In the first part of this volume the nitrogen containing compounds of molybdenum are described The Mo N system shows that Mo_3N_2 and Mo_3N are the stable nitrides Mo metal dissolves nitrogen to some extent but only at high temperatures To get better insight into the reactions between nitrogen and molybdenum the solubility diffusion adsorption and desorption phenomena and ion bombardment are included in the section of the Mo N system Mo_3N has a large range of homogeneity toward lower nitrogen concentrations The black α hexagonal Mo_3N has only a narrow range of homogeneity In addition some molybdenum compounds containing nitrogen and oxygen are known The second part contains a full description of the compounds of molybdenum with fluorine The fluorides MoF_n with $n = 2$ are metastable while those with $n = 3$ to 6 are stable and have been observed in the Mo F system Pure MoF_3 can exist without traces of oxygen in contrast to earlier assumptions MoF_3 was unambiguously prepared and characterized in 1957 Its crystal structure is still unknown MoF_3 is often contaminated with the oxide fluoride MoOF_2 and it is difficult to remove Even small amounts affect the properties of MoF_3 MoF_3 which is liquid at room temperature and solidifies to a plastic crystal modification below ca 17°C is the most investigated of all the molybdenum fluorides Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career Tin: Diorganotin-oxygen compounds with R_2Sn , $\text{RR}'\text{Sn}$, or R-Sn units and with identical or different oxygen-bonded groups, 1988 **Water Chemistry of Nuclear Reactor Systems** British Nuclear Energy Society, 1978

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