

## Chapter Test B

### Chapter 1: Matter and Change

From 1 to 10, copy the words in the boxes in the order in which they have been numbered, leaving a space for each number before you write.

1. Matter is composed of tiny units that always  
a. moving together and that are permanent  
b. have different characteristics  
c. have properties that are shared by many substances and substances  
d. have properties that are shared by many substances and substances

2. The law of constant composition is a rule that says  
a. substances have different properties  
b. substances have different properties  
c. substances have different properties  
d. substances have different properties

3. The law of multiple proportions is a rule that says  
a. substances have different properties  
b. substances have different properties  
c. substances have different properties  
d. substances have different properties

4. The law of conservation of mass is a rule that says  
a. mass is lost in a reaction  
b. mass is gained in a reaction  
c. mass is lost in a reaction  
d. mass is gained in a reaction

5. If 100 g of water are combined with 100 g of ethanol  
a. 100 g      100 g      100 g      100 g  
b. 100 g      100 g      100 g      100 g  
c. 100 g      100 g      100 g      100 g  
d. 100 g      100 g      100 g      100 g

6. All matter consists of the same elements  
a. changing the elements  
b. changing the elements  
c. changing the elements  
d. changing the elements

7. In 1911, Ernest Rutherford and his team found the following properties of atoms  
a. electrons are negatively charged  
b. electrons are positively charged  
c. electrons are positively charged  
d. electrons are positively charged

8. Rutherford proposed the nuclear model of the atom. This model suggested the atom  
a. has a central nucleus and electrons orbit the nucleus  
b. has a central nucleus and electrons orbit the nucleus  
c. has a central nucleus and electrons orbit the nucleus  
d. has a central nucleus and electrons orbit the nucleus

9. The only property of atoms that is not shared by all atoms is  
a. mass and volume  
b. number of protons  
c. number of neutrons  
d. number of electrons

10. The first requirement of the  $\text{H}_2\text{O}$  molecule is that it  
a. has two hydrogen atoms  
b. has one oxygen atom  
c. has two oxygen atoms  
d. has three hydrogen atoms

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### **Table of Contents Assessment Chapter Test B**

1. Understanding the eBook Assessment Chapter Test B
  - The Rise of Digital Reading Assessment Chapter Test B
  - Advantages of eBooks Over Traditional Books
2. Identifying Assessment Chapter Test B
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Assessment Chapter Test B
  - User-Friendly Interface
4. Exploring eBook Recommendations from Assessment Chapter Test B
  - Personalized Recommendations
  - Assessment Chapter Test B User Reviews and Ratings
  - Assessment Chapter Test B and Bestseller Lists

5. Accessing Assessment Chapter Test B Free and Paid eBooks
  - Assessment Chapter Test B Public Domain eBooks
  - Assessment Chapter Test B eBook Subscription Services
  - Assessment Chapter Test B Budget-Friendly Options
6. Navigating Assessment Chapter Test B eBook Formats
  - ePUB, PDF, MOBI, and More
  - Assessment Chapter Test B Compatibility with Devices
  - Assessment Chapter Test B Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Assessment Chapter Test B
  - Highlighting and Note-Taking Assessment Chapter Test B
  - Interactive Elements Assessment Chapter Test B
8. Staying Engaged with Assessment Chapter Test B
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Assessment Chapter Test B
9. Balancing eBooks and Physical Books Assessment Chapter Test B
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Assessment Chapter Test B
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Assessment Chapter Test B
  - Setting Reading Goals Assessment Chapter Test B
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Assessment Chapter Test B
  - Fact-Checking eBook Content of Assessment Chapter Test B
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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