

## SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

- What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water
- Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
- According to the graph, the solubility of any substance changes as temperature changes.
- List the substances whose solubility decreases as temperature increases.  $\text{NH}_3$  and  $\text{Ce}_2(\text{SO}_4)_3$
- Which substance is least affected by temperature changes?  $\text{NaCl}$
- How many grams of ammonium chloride ( $\text{NH}_4\text{Cl}$ ) at  $50^\circ\text{C}$ ? 50g
- $\text{NaCl}$  and  $\text{KClO}_3$  have the same solubility at approximately  $78^\circ\text{C}$ .
- Which compound is least soluble in water at  $10^\circ\text{C}$ ?  $\text{KClO}_3$
- How many grams of  $\text{KNO}_3$  can be dissolved at  $50^\circ\text{C}$ ? 80g
- Are the following solutions unsaturated, saturated, or supersaturated?
  - 45g of  $\text{NaNO}_3$  in 100 g of water at  $30^\circ\text{C}$ . saturated
  - 60g of  $\text{KClO}_3$  in 100 g of water at  $90^\circ\text{C}$ . supersaturated
- How many grams of sodium chloride,  $\text{NaCl}$  are required to saturate 100 grams of water at  $100^\circ\text{C}$ ? 40g
- How many grams of  $\text{NaNO}_3$  are required to saturate 100 grams of water at  $75^\circ\text{C}$ ? 140g
- How many grams of  $\text{KI}$  will saturate water at  $20^\circ\text{C}$ ? 33g
- At what temperature would 25g of potassium chlorate ( $\text{KClO}_3$ ) dissolve? 60°C
- At what temperature would 60g of  $\text{NH}_4\text{Cl}$  dissolve? 70°C
- 89 g  $\text{NaNO}_3$  is prepared at  $30^\circ\text{C}$ .
  - Will all of the salt dissolve? No
  - What mass of  $\text{NaNO}_3$  will dissolve at this temperature? 95g
- If 50 grams of  $\text{NH}_4\text{Cl}$  is dissolved at  $50^\circ\text{C}$ , how many additional grams  $\text{NH}_4\text{Cl}$  would be needed to make the solution saturated at  $80^\circ\text{C}$ ? 15g
- At  $50^\circ\text{C}$ , how many grams of  $\text{KNO}_3$  will dissolve? 80g
- At  $70^\circ\text{C}$ , how many grams of cerium (III) sulfate ( $\text{Ce}_2(\text{SO}_4)_3$ ) dissolve? 5g
- Determine if each of the following is unsaturated, saturated, or supersaturated.

a. 55g of $\text{NH}_3$ at $20^\circ\text{C}$ <u>supersaturated</u>	f. 78g of $\text{NaNO}_3$ at $10^\circ\text{C}$ . <u>saturated</u>
b. 10g of $\text{Ce}_2(\text{SO}_4)_3$ at $10^\circ\text{C}$ <u>unsaturated</u>	g. 145g of $\text{NaNO}_3$ at $80^\circ\text{C}$ . <u>saturated</u>
c. 110g of $\text{KNO}_3$ at $60^\circ\text{C}$ . <u>supersaturated</u>	h. 35g of $\text{NaCl}$ at $100^\circ\text{C}$ . <u>unsaturated</u>
d. 65g of $\text{NH}_4\text{Cl}$ at $80^\circ\text{C}$ . <u>saturated</u>	
e. 12g of $\text{NH}_3$ at $90^\circ\text{C}$ . <u>supersaturated</u>	

# Answer Key To Lab Solubility Data Sheet

**Carolina Biological Supply Company**

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**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran,2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

**Science Educator's Guide to Laboratory Assessment** Rodney L. Doran,2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab

**Basic Laboratory Principles in General Chemistry** Fitzgerald B. Bramwell,1990      **Chemical Interactions** ,2005      **Title**

**List of Documents Made Publicly Available** U.S. Nuclear Regulatory Commission,1985      **Laboratory Manual for Principles of General Chemistry** J. A. Beran,2013-12-12 A two term manual for General Chemistry This supplementary manual focuses on chemical principles and techniques The Laboratory Manual for Principles of General Chemistry tenth edition provides a broad scope of experiments coupled with a clear layout for ease of use The manual delivers material for two or three course terms It also assists chemistry students in knowing how to time various techniques in the lab environment The companion manual is organized into topic sections such as Chemical and Physical Properties Atomic and Molecular Structure Gases and Solutions

**A Laboratory Manual of Pharmacy for Veterinary Students** Bernard Victor Alfredson,1943

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The experiments in this manual are designed in a discovery format and the majority require only small quantities of reagents      The Petroleum Engineer ,1947      Regulatory and Technical Reports (abstract Index Journal). U.S. Nuclear Regulatory Commission. Regulatory Publications Branch,1988      Carolina Science and Math Carolina Biological Supply Company,2003  
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