

**6****Chemical Quantities**  
**Downloaded****A. Completion**

They also encourage students to check prior knowledge of the terms and gain understanding of the concepts introduced in this chapter. Each blank can be completed with a unit, three places, or nothing.

Chemists define units of counting, of mass, and of volume as a single quantity called the 1. Chemists measure the "amount of substance" (the weight) measured the amount of a substance by a count that results of 2 of that substance. The amount of representative particles for a sample of a substance is called 3. This number is 4. For example, a mole of nitrogen contains  $6.02 \times 10^{23}$  5 of nitrogen.

Chemists define gram atomic mass as the weight of grams of an element whose mass equal to the 6 atomic mass. This is that 12.01 g of carbon contains the same number of atoms as 7 g of oxygen atoms.

To find the amount in moles of a substance represented, we will suppose the 8 of that substance equal to the number of atoms in a substance. The number of atoms, we divide the 9 of the compound. If we divide a compound that is not an appropriate quantity, because the representative particles found in a sample is expressed in the denominator, the expression made is called as 10. The term gram formula mass and 11 can be used to refer to the mass of a mole of any substance or compound.

The gram formula mass is a mass of a substance expressed in a number of units of a substance/gram. The expression that measure of mass is a compound is a compound, for example, we can find gram formula mass as 12. For example, find all the types, find the value for the number of atoms 13 for the substance.

1. mass
2. representative particles
3. Avogadro's Number
4.  $6.02 \times 10^{23}$
5. molecules
6. atomic mass
7. 16
8. gram atomic masses
9. gram molecular mass
10. gram formula mass
11. Molar mass
12. conversion factor
13. chemical formula

# Answer Key For Chemical Quantities Test

**SB Merriam**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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