

CHAPTER 3**CHAPTER ASSESSMENT****MATTER AND CHANGE****Reviewing Vocabulary**

Match the definition in Column A with the term in Column B.

Column A

1. The outermost electrons of an atom
2. The center of an atom
3. Hot, highly ionized, electrically conducting gas
4. The smallest particle of an element that retains that element's characteristics
5. The change from a gas to a liquid
6. A homogeneous mixture
7. The change of one or more substances into other substances
8. The attractive force between two ions of opposite charge
9. A tiny particle with a negative electrical charge
10. The average of the mass numbers of an element's isotopes
11. The spontaneous process through which unstable nuclei emit radiation

Column B

- a. atom
- b. atomic mass
- c. chemical reaction
- d. condensation
- e. electron
- f. ionic bond
- g. solution
- h. nucleus
- i. plasma
- j. valence electrons
- k. radioactive decay

In the space at the left, write the word or phrase in parentheses that makes the statement correct.

12. The combined number of protons and neutrons is the (atomic number, mass number).
13. An (energy level, isotope) represents the area in an atom where an electron is most likely to be found.
14. (An ionic bond, A covalent bond) is the attraction of two atoms for a shared pair of electrons that holds the atoms together.
15. An atom that gains or loses an electron is a charged particle called (an ion, a molecule).
16. (An element, A compound) is a substance that cannot be broken down into simpler substances.
17. A tiny particle that has mass and a positive electrical charge is a (neutron, proton).
18. (Acids, Bases) are solutions characterized by the formation of hydroxide ions (OH^-).
19. The process of changing from a liquid into a gas is called (evaporation, sublimation).

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Michael Brown

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