

# Acids & Bases: Assessment Review

Name \_\_\_\_\_

*Skill – The student will be able to identify the acid, base, conjugate acid, and conjugate base in an acid – base reaction.*

1. Write a chemical equation showing how  $\text{HCO}_3^-$  behaves as an acid when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

2. Write a chemical equation showing how ammonia ( $\text{NH}_3$ ) behaves as a base when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

*Skill – The student will be able to calculate with the  $[\text{H}^+]$  or  $[\text{OH}^-]$  from the ion product ( $K_w$ ) or pH from pOH (a one-step pH block calculation).*

3. Recreate the calculation flowchart on the diagram provided. (Be careful...it's been flipped!)



4. What is  $[\text{OH}^-]$  if a solution has  $[\text{H}^+] = 7.1 \times 10^{-4} \text{ M}$ ? Is the solution acidic, basic, or neutral?

5. Calculate  $[\text{H}^+]$  in a solution if  $[\text{OH}^-] = 5.1 \times 10^{-11} \text{ M}$ . Is the solution acidic, basic, or neutral?

6. Calculate the  $[\text{H}^+]$  in a solution that shows a pH of 11.75. Is the solution acidic, basic, or neutral?

7. What is the pOH of a solution if  $[\text{OH}^-] = 5.0 \times 10^{-4} \text{ M}$ ? Is the solution acidic, basic, or neutral?

# Assessment Acids And Bases Answer Key

**John Eichinger**



## **Assessment Acids And Bases Answer Key:**

**Fluids and Electrolytes** Catherine Paradiso, 1999 This book is designed to prepare students for classes or NCLEX by providing a comprehensive outline review of this particularly difficult area of study Fluids and Electrolytes offers extensive self testing that includes answer keys with rationale for correct and incorrect responses as well as a comprehensive outline review and chapter study questions and a comprehensive examination at the end of the book Each question is categorized according to the components of the National Council of State Boards of Nursing Licensing Examinations NCLEX Includes FREE disk containing more NCLEX questions

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**Fluid, Electrolyte, and Acid-Base Imbalances** Allison

Hale, Mary Jo Hovey, 2013-10-29 Here's all of the crucial coverage you need to succeed in class and confidently prepare for your classroom exams and the NCLEX Easy to follow outlines focus on the information essential to make this challenging subject more manageable

Chemistry for the Utterly Confused John T. Moore, Richard H. Langley, 2007-06-01 Banish bafflement in this tough subject From formulas and lab techniques to the periodic table Chemistry for the Utterly Confused focuses on the areas of maximum confusion and breaks down the most difficult chemistry topics into easy to understand concepts This invaluable guide also teaches problem solving skills you need to master this imposing subject Whether you're in high school in college or simply brushing up on chemistry knowledge this fun easily accessible book will make understanding chemistry a breeze

**Critical Care Manual of Clinical Nursing Procedures** Suzanne Bench, Nicki Credland, Chris Hill, 2024-10-30 Critical Care Manual of Clinical Nursing Procedures The second edition of Critical Care Manual of Clinical Nursing Procedures is a practical overview of essential procedures for the care of critically ill patients Beginning with chapters outlining the current scope of critical care the book adopts a systematic stage by stage approach from admission to discharge At each stage it provides insights into physiology key procedures and the relevant evidence base Now fully updated to incorporate the latest research and best practices this volume is poised to remain an indispensable resource for the next generation of critical care providers Readers of the second edition will find In depth beat by beat analysis of key procedures in critical care Interventions underpinned by the latest evidence Content aligned with the National Critical Care Competency Framework and endorsed by the British Association of Critical Care Nurses Critical Care Manual of

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*Critical Care Review for Nurses* Sharyn A. Tremblay, 1984-04      **Chemical Interactions**, 2005      Exercises in Second Year Chemistry William Henry Chapin, 1944

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