

## Chemistry

## Stoichiometry Reflect #2

Learning Target: I can use stoichiometry to calculate amounts.

Name Ms. Hettie

1a. Balance this equation



1b. Determine how many moles of ammonium oxide was used if the student weighed out 55.2 g of it.

$$55.2 \text{ g (NH}_4\text{)}_2\text{O} \left( \frac{1 \text{ mol A}}{52.10 \text{ g}} \right) = 1.06 \text{ moles (NH}_4\text{)}_2\text{O}$$

1c. Determine the amount of moles of ammonium nitride produced in this reaction.

$$1.06 \text{ moles A} \left( \frac{2 \text{ mol B}}{3 \text{ mol A}} \right) = 0.706 \text{ moles (NH}_4\text{)}_3\text{N}$$

1d. Determine how many grams of ammonium nitride produced in this reaction.

$$0.706 \text{ moles} \left( \frac{85.16 \text{ g}}{1 \text{ mol B}} \right) = 48.16 \text{ g (NH}_4\text{)}_3\text{N}$$

2. Using the following chemical reaction:  $2 \text{ Hg} + \text{Cu}(\text{NO}_3)_2 \rightarrow 2 \text{ HgNO}_3 + \text{Cu} \xrightarrow{A}$   
and, knowing that 61.6 g of copper was produced in this experiment, how many molecules of mercury nitrate was also produced?

$$61.6 \text{ g Cu} \left( \frac{1 \text{ mol A}}{63.55 \text{ g}} \right) \left( \frac{2 \text{ mol B}}{1 \text{ mol A}} \right) \left( \frac{6.02 \times 10^{23} \text{ mc}}{1 \text{ mol B}} \right) = 1.17 \times 10^{24} \text{ mc HgNO}_3$$

3. Challenge...

What is the percent yield of the following reaction if 12.8 g of copper (II) chloride was used in the lab, and 7.75 g of aluminum chloride was measured at the end of the experiment? (hint: you must calculate the theoretical yield of aluminum chloride first!!)



$$12.8 \text{ g A} \left( \frac{1 \text{ mol A}}{134.45 \text{ g}} \right) \left( \frac{2 \text{ mol B}}{3 \text{ mol A}} \right) \left( \frac{133.33 \text{ g}}{1 \text{ mol B}} \right) = 8.46 \text{ g AlCl}_3 \text{ theoretical}$$

$$\begin{aligned} \text{yield} &= 7.75 \text{ g} \\ &\quad 8.46 \text{ g} \\ &= 91.4\% \end{aligned}$$

Show work on back

# Answer Key To Mixed Stoichiometry Practice

**R. Bruce King**

## Answer Key To Mixed Stoichiometry Practice:

**Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998     *Basics of Analytical Chemistry and Chemical Equilibria* Brian M. Tissue, 2013-07-22 Enables students to progressively build and apply new skills and knowledge. Designed to be completed in one semester this text enables students to fully grasp and apply the core concepts of analytical chemistry and aqueous chemical equilibria. Moreover the text enables readers to master common instrumental methods to perform a broad range of quantitative analyses. Author Brian Tissue has written and structured the text so that readers progressively build their knowledge beginning with the most fundamental concepts and then continually applying these concepts as they advance to more sophisticated theories and applications. **Basics of Analytical Chemistry and Chemical Equilibria** is clearly written and easy to follow with plenty of examples to help readers better understand both concepts and applications. In addition there are several pedagogical features that enhance the learning experience including Emphasis on correct IUPAC terminology. You Try It spreadsheets throughout the text challenging readers to apply their newfound knowledge and skills. Online tutorials to build readers skills and assist them in working with the text's spreadsheets. Links to analytical methods and instrument suppliers. Figures illustrating principles of analytical chemistry and chemical equilibria. End of chapter exercises. **Basics of Analytical Chemistry and Chemical Equilibria** is written for undergraduate students who have completed a basic course in general chemistry. In addition to chemistry students this text provides an essential foundation in analytical chemistry needed by students and practitioners in biochemistry environmental science chemical engineering materials science nutrition agriculture and the life sciences     **A Stoichiometry Unit** David Callaghan, 2004

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**Forest Microbiome: Dynamics and Interactions in the**

**Anthropocene Era** Amit Roy,Amrita Chakraborty,Shulin He,Fred O. Asiegbu,Teresa Ann Coutinho,Antonio Castellano-Hinojosa,2025-12-10 Forests encompass around 40.6 million km<sup>2</sup> of the Earth's surface making them a vital component of our ecosystem. Not only do they contribute to reducing carbon footprints and influencing biogeochemical cycles but they also play a crucial role in mitigating environmental changes during the Anthropocene era. However, forests face numerous challenges including drought, forest fires, warming temperatures, increased carbon dioxide levels, pest and pathogen outbreaks, and nitrogen deposition. These challenges significantly impact microbial communities within the forest ecosystem. Drought, for instance, negatively affects forest resilience, leading to forest fires and outbreaks of pests and pathogens that have a profound influence on microbial processes. Forest microbiomes are essential for maintaining the health and functioning of the forest ecosystem by engaging in balanced interactions with their tree hosts. Disruptions to this delicate balance can have adverse effects on host fitness. Microbes also play a crucial role in nutrient availability and recycling carbon dioxide sequestration and disease resistance in trees. As a result, they have a significant impact on ecosystem dynamics and overall forest health. Additionally, symbiotic relationships within forest pests give them an advantage in adapting to challenges and seizing opportunities during the Anthropocene era. Understanding the intricate relationship between forests, forest inhabitants, and microbes is therefore critical for effective forest management, conservation efforts, and mitigating ongoing environmental changes. This Research Topic aims to address microbial diversity and its functional role in maintaining forest health and ecosystem dynamics during the Anthropocene. Moreover, it seeks to provide insights into novel approaches for studying the composition of the ecosystem's microbiome. We will also examine how the microbiome responds to environmental changes such as disease and pest outbreaks, climate change, drought, and their potential in forest management and conservation efforts. The scope of this Research Topic is expansive and not limited to the following subjects:

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Brady,John R. Holum,1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving *Kirk-Othmer Encyclopedia of Chemical Technology, Carbon and Graphite Fibers to Chlorocarbons and Chlorohydrocarbons-CSUB 1/SUB* Kirk-Othmer,1993-01-22

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