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1. In a cathode ray tube:
 - electrons pass from the anode to the cathode.
 - electrons pass from the cathode to the anode.
 - protons pass from the anode to the cathode.
 - protons pass from the cathode to the anode.

Ans: B Category: Medium Section: 2.2
2. The scientist who determined the magnitude of the electric charge of the electron was:

A) John Dalton	D) Henry Moseley
B) Robert Millikan	E) R. Chang
C) J. J. Thomson	

Ans: B Category: Easy Section: 2.2
3. When J. J. Thomson discovered the electron, what physical property of the electron did he measure?

A) its charge, e	D) its mass, m
B) its charge-to-mass ratio, e/m	E) its atomic number, Z
C) its temperature, T	

Ans: B Category: Easy Section: 2.2
4. Which of the following scientists developed the nuclear model of the atom?

A) John Dalton	D) Henry Moseley
B) Robert Millikan	E) Ernest Rutherford
C) J. J. Thomson	

Ans: E Category: Easy Section: 2.2
5. Rutherford's experiment with alpha particle scattering by gold foil established that:
 - protons are not evenly distributed throughout an atom.
 - electrons have a negative charge.
 - electrons have a positive charge.
 - atoms are made of protons, neutrons, and electrons.
 - protons are 1840 times heavier than electrons.

Ans: A Category: Medium Section: 2.2
6. Atoms of the same element with different mass numbers are called:
 - ions.
 - neutrons.
 - allotropes.
 - chemical families.
 - isotopes.

Ans: E Category: Easy Section: 2.3

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