

Identifying the 5 Types of Chemical Reactions

Identify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)

Name of Reaction Type

1. $2 \text{Na} + \text{Cl}_2 \rightarrow 2 \text{NaCl}$ EX: A B → AB (A & B combine)	Ex: synthesis
2. $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$	
3. $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$	
4. $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$	
5. $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$	
6. $\text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$	
7. $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$	
8. $\text{C}_6\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$	
9. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$	
10. $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2 \text{Ag}$	
11. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$	

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