

Topic:

Water and Aqueous Systems

1. In the absence of any external forces, the shape of a drop of water in zero-gravity is best described by which of the following?
- surface tension
 - density
 - viscosity
 - buoyancy
2. The average molar mass of hydrous hematite (Fe_2O_3) is greater than that of anhydrous (Fe_2O_3) which is greater than the molar mass of the hydrous hematite-hydroxide ($\text{Fe}_2\text{O}_3 \cdot \text{H}_2\text{O}$) which is the sum of the molar mass of anhydrous hematite and water.
- water
 - anhydrous
 - hydrous hematite
3. Cobalt chloride (CoCl_2) is a blue-white solid and a hygroscopic solid and can be reduced to green translucent crystals. It dissolves in water to form a pinkish-red solution. Cobalt chloride ($\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$) is the product of hydration of cobalt chloride. Which of the following best describes cobalt chloride?
- hydrous
 - anhydrous
 - crystalline
 - amorphous

Answer Key Of Water Aqueous Systems

J Elliott



Answer Key Of Water Aqueous Systems:

Selected Water Resources Abstracts ,1988 **Selected Water Resources Abstracts** ,1988 Thermodynamics of Solutions Eli Ruckenstein,Ivan L. Shulgin,2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade Even though they involve different topics and different systems they have something in common which can be considered as the signature of the present book First these papers are concerned with difficult or very nonideal systems i e systems with very strong interactions e g hyd gen bonding between components or systems with large differences in the partial molar v umes of the components e g the aqueous solutions of proteins or systems that are far from normal conditions e g critical or near critical mixtures Second the conventional th modynamic methods are not sufficient for the accurate treatment of these mixtures Last but not least these systems are of interest for the pharmaceutical biomedical and related ind tries In order to meet the thermodynamic challenges involved in these complex mixtures we employed a variety of traditional methods but also new methods such as the fluctuation t ory of Kirkwood and Buff and ab initio quantum mechanical techniques The Kirkwood Buff KB theory is a rigorous formalism which is free of any of the proximations usually used in the thermodynamic treatment of multicomponent systems This theory appears to be very fruitful when applied to the above mentioned difficult systems

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