

Name Key
Specific Heat Worksheet

$C_p = q/m\Delta T$, where q = heat energy, m = mass, and T = temperature

1. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the heat capacity of iron.

$$1086.75 = 15.75g \cdot 150 \cdot C_p$$

$$1086.75 = 2362.5 C_p$$

$$C_p = .46 \text{ J/g}^\circ\text{C}$$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$q = C_p \cdot m \cdot \Delta T \quad .90(10)(33) =$$

$$297 \text{ J}$$

$$\frac{55}{22} = 33$$

3. To what temperature will a 50.0 g piece of glass raise if it absorbs 5275 joules of heat and its heat capacity is 0.50 J/g°C? The initial temperature of the glass is 20.0°C.

$$5275 = .50(50)\Delta T$$

$$\Delta T = 211$$

$$T_f = \frac{211}{+20} = 231$$

$$231^\circ\text{C}$$

4. Calculate the heat capacity of a piece of wood if 1500.0 g of the wood absorbs 6.75×10^4 joules of heat, and its temperature changes from 32°C to 57°C.

$$67500 = C_p \cdot 1500 \cdot 25^\circ\text{C}$$

$$67500 = 37500 C_p$$

$$1.8 \text{ J/g}^\circ\text{C}$$

5. 100.0 mL of 4°C water is heated until its temperature is 37°C. If the specific heat of water is 4.18 J/g°C, calculate the amount of heat needed to cause this rise in temperature.

$$Q = 100g \cdot 4.18 \cdot 33^\circ\text{C}$$

$$= 13794$$

$$13794 \text{ J}$$

$$13794$$

6. 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the heat capacity of mercury.

$$455 = C_p \cdot 25g \cdot 130$$

$$455 = 3250 C_p$$

$$.14 = C_p$$

7. What is the Specific heat capacity of silver metal if 55.00 g of the metal absorbs 198 joules of heat and the temperature rises 15.0°C?

$$198 = C_p \cdot 55g \cdot 15^\circ\text{C}$$

$$198 = 825 C_p$$

$$C_p = .24 \text{ J/g}^\circ\text{C}$$

8. What is the change in temperature of 150.0 g of chloroform if it absorbs 1000.0 joules of heat, and the specific heat of chloroform is 0.96 J/g°C?

$$1000 = .96(150g)\Delta T$$

$$1000 = 144 \Delta T$$

$$6.94 = \Delta T$$

$$6.9^\circ\text{C}$$

Answers:

- $C_p = .46 \text{ J/g}^\circ\text{C}$
- $Q = 297 \text{ J}$
- $T_{\text{final}} = 231^\circ\text{C}$
- $C_p = 1.8 \text{ J/g}^\circ\text{C}$
- $Q = 13800 \text{ J}$ 13794
- $C_p = .14 \text{ J/g}^\circ\text{C}$
- $C_p = .24 \text{ J/g}^\circ\text{C}$
- $\Delta T = 6.9^\circ\text{C}$

Answer Key For Specific Heat

O García

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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AQA GCSE Physics for Combined Science: Trilogy 9-1 Student Book (GCSE Science 9-1) Sandra Mitchell,Charles Golabek,2021-09-20 Exam Board AQA Level Subject GCSE Combined Science Trilogy First teaching September 2016 First exams June 2018 AQA approved Holt Science and Technology Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **English Mechanic and Mirror of Science and Art** ,1881 **The Right to Literacy in Secondary**

Schools Suzanne Plaut,2009 This inspirational resource challenges educators to view adolescent literacy as a civil right that enables students to understand essential content and to develop as independent learners Edited by the Vice President of Education at the Public Education Business Coalition PEBC a nonprofit that is nationally known for its expertise in literacy and professional development the book is a call to action and a practical guide for reform minded schools and districts and for teachers seeking to help all adolescent learners achieve at high levels It is replete with vivid illustrations of exemplary classroom practice across all content areas It also offers important frameworks to help teachers implement those practices in their own schools Perfect for professional learning communities study groups and individual teachers this book Offers schools and districts inspiration and guidance for undertaking a whole school inquiry about adolescent learning thinking and literacy Shows the specific role that literacy plays in each of the core content areas math science social studies and language arts Includes proven tips for secondary and middle school teachers based on the work of teachers in classrooms where

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 - Chapter 4: Answer Key For Specific Heat in Specific Contexts
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