

Ionic Compounds Practice

1. Name these compounds:



2. Write the chemical formula for each of the following:

(a) potassium iodide

(b) aluminum oxide

(c) calcium chloride

(d) zinc bromide

(e) sodium fluoride

(f) sodium oxide

3. Name these compounds:



4. Write the chemical formula for each of the following:

(a) K_2O

(b) Ca_3O_2 Boron trioxide

(c) K_2O carbon dioxide

(d) Na_2O_2

(e) Ca_3O_2 boron trioxide

(f) K_2O carbon dioxide

(g) K_2O_2 boron trioxide

(h) Ca_3O_2 boron trioxide

(i) K_2O_2 carbon dioxide

(j) K_2O_2 boron trioxide

(k) Ca_3O_2 boron trioxide

(l) K_2O_2 carbon dioxide

(m) K_2O_2 boron trioxide

(n) Ca_3O_2 boron trioxide

(o) K_2O_2 carbon dioxide

(p) K_2O_2 boron trioxide

(q) Ca_3O_2 boron trioxide

(r) K_2O_2 carbon dioxide

(s) K_2O_2 boron trioxide

(t) Ca_3O_2 boron trioxide

(u) K_2O_2 carbon dioxide

(v) K_2O_2 boron trioxide

(w) Ca_3O_2 boron trioxide

(x) K_2O_2 carbon dioxide

Answer Key For Making Models Of Compounds

G Orfield

Answer Key For Making Models Of Compounds:

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