

7. Alkanols can be manufactured from alkenes by the initial reaction of alkenes with

- bromine in tetrachloromethane.
- concentrated tetraxosulphate(VI) acid.
- aqueous potassium tetraxomanganate(VII).
- sodium hydroxide solution.

8. Which of the following statements about the standard hydrogen electrode is not correct?

- The hydrogen gas is at a pressure of 1 atmosphere.
- A solution containing 1 mol dm^{-3} of H^+ ions is used.
- A platinum electrode is used.
- The temperature is kept at 20°C .

9. If 60 g of M combines with 24 g of oxygen, what would the empirical formula of the oxide be?
(O = 16.0, M = 120)

- MO
- MO₂
- MO₃
- MO₄

10. The products of the electrolysis of dilute sodium chloride using carbon electrodes are

<ol style="list-style-type: none"> chlorine and oxygen and chlorine and hydrogen and 	<ol style="list-style-type: none"> anode anode cathode cathode 	<ol style="list-style-type: none"> sodium. hydrogen. hydrogen. oxygen.
--	--	--

11. Determine the quantity of electricity used when a current of 0.20 amperes is passed through an electrolytic cell for 60 minutes.

- 12 C
- 120 C
- 360 C
- 720 C

12. Oxochlorate(I) acid is used as a bleaching agent because it is

- a weak acid.
- a reducing agent.
- an oxidizing agent.
- a strong acid.

13. The IUPAC name for $\text{CH}_3\text{CHCH}_3\text{CHClCHCH}_3\text{CH}_2\text{CH}_3$ is

- 2, 4 - dimethyl - 3 - chlorohexane.
- 3, 5 - dimethyl - 4 - chlorohexane.
- 4 - chloro - 3, 5 - dimethylhexane.
- 3 - chloro - 2, 4 - dimethylhexane.

14. A colourless gas with a pungent smell is evolved when dilute hydrochloric acid is added to a sample of a salt. The gas evolved could turn

- acidified $\text{K}_2\text{Cr}_2\text{O}_7$ solution colourless.
- acidified KMnO_4 solution colourless.
- $\text{Fe}(\text{NO}_3)_2$ solution green.
- $\text{Pb}(\text{NO}_3)_2$ paper black.

15. If 5.0 g of marble reacts with 25.0 cm^3 hydrochloric acid, which of the following combinations has the fastest reaction rate?

- Marble chips and 2.0 mol dm^{-3} $\text{HCl}_{(\text{aq})}$
- Marble chips and 2.5 mol dm^{-3} $\text{HCl}_{(\text{aq})}$
- Powdered marble and 2.5 mol dm^{-3} $\text{HCl}_{(\text{aq})}$
- Powdered marble and 2.0 mol dm^{-3} $\text{HCl}_{(\text{aq})}$

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