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SERVICE MANUAL

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web solution if we compare the time that has passed to the isotope's half life we note that 44.0 s is exactly 4 half lives so using the previous equation n/4 substituting and solving results in the following less than one third of a gram of ^{20}F remains skill building exercise the half life of ^{44}Ti is 60.0 y

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web solution 7.2 2 4.3 half lives 1.2 0 125 the amount remaining after 3 half lives $100.0 \text{ g} \times 0.125 = 12.5 \text{ g}$ remaining problem 2 pd 100 has a half life of 3.6 days if one had 6.02×10^{23} atoms at the start how many atoms would be present after 20.0 days solution 20.0 3.6 5.56 half lives

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web radioactive isotopes are used for blood flow monitoring cancer treatment paper mills carbon dating and smoke alarms each isotope used in these applications has a characteristic half life