

## Chemical Kinetics

- 1) Molecular hydrogen can be made from methane gas by the reaction below. How is the rate of disappearance of  $\text{CH}_4$  related to the rate of appearance of  $\text{H}_2$ ?



- A) \_\_\_\_\_  
 B) \_\_\_\_\_  
 C) \_\_\_\_\_  
 D) none of these

Answer: B

- 2) For the reaction shown below, what is the relationship between the rate of formation of  $\text{Br}_2$  and the rate of reaction of  $\text{Br}^-$  over the same time period?



- A) rate of formation of  $\text{Br}_2 = 1/2 \times$  rate of reaction of  $\text{Br}^-$   
 B) rate of formation of  $\text{Br}_2 = 1/4 \times$  rate of reaction of  $\text{Br}^-$   
 C) rate of formation of  $\text{Br}_2 = 2 \times$  rate of reaction of  $\text{Br}^-$   
 D) rate of formation of  $\text{Br}_2 = 2 \times$  rate of reaction of  $\text{Br}^-$

Answer: A

- 3) The reaction that occurs in a Breathalyzer, a device used to determine the alcohol level in a person's bloodstream, is given below. If the rate of appearance of  $\text{Cr}_2(\text{SO}_4)_3$  is  $1.34 \text{ mol/min}$  at a particular moment, what is the rate of disappearance of  $\text{C}_2\text{H}_5\text{O}$  at that moment?



- A)  $0.413 \text{ mol/min}$   
 B)  $0.826 \text{ mol/min}$   
 C)  $1.86 \text{ mol/min}$   
 D)  $3.72 \text{ mol/min}$

Answer: C

# 1996 Ap Chem Multiple Choice Answers

**Xiang Bo Wang**



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