

# The Person I Admire The Most Is My Mother Essay

Crafting an essay on the person I admire the most, my mother, is no easy task. It involves navigating through a plethora of emotions, memories, and experiences that have shaped my admiration for her. The challenge lies not only in capturing the essence of her character but also in conveying the depth of my feelings in a coherent and engaging manner.

First and foremost, there's the emotional aspect. Describing the love and admiration for someone so significant in one's life requires a delicate balance of sentimentality without becoming overly sentimental. It's about sharing personal anecdotes, moments of strength, and instances of unwavering support while maintaining a level of objectivity to make the narrative relatable to a broader audience.

Then comes the challenge of articulating the qualities that make my mother admirable. It's not just about listing virtues but providing vivid examples and illustrations that showcase her kindness, resilience, and wisdom. This requires a keen attention to detail, selecting the right words to evoke a sense of appreciation in the reader.

Additionally, organizing the essay poses its own set of challenges. Finding the right structure to present a coherent and flowing narrative is crucial. Starting with a compelling introduction, weaving through various aspects of her personality and impact, and concluding with a meaningful reflection requires careful planning and execution.

The difficulty is compounded by the fear of falling into clichés or generic descriptions. It's crucial to avoid overly used phrases and expressions while ensuring that the essay remains genuine and unique to my personal experience. Striking this balance demands a thoughtful and creative approach to language and expression.

In the end, tackling an essay on the person I admire the most, my mother, requires a blend of emotional vulnerability, precise language, and strategic organization. It's a journey through memories and sentiments that demands both introspection and effective communication skills to do justice to the profound impact she has had on my life.

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# Who I Admire Essays

**Francis Bacon**



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web consider the thermochemical equation below  $4\text{NO(g)} + 6\text{H}_2\text{O(l)} + 4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 11\text{H}_2\text{O(l)} + 7\text{N}_2\text{(g)}$  a is this reaction endothermic or exothermic b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of  $\text{H}_2$

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web sep 5 2023 3 carry out a thermochemical calculation on all these species at their optimized geometries 4 sum up the enthalpies free energy of the products and the reagents keeping into account the

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore  $\Delta H$  must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the  $\Delta H$  per mole of  $\text{O}_2$  and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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web thermochemistry practice worksheet answer key what is the heat change when 472 g of carbon reacts with excess  $O_2$  according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat  
 $2S_3O_2 + 2SO_3 \rightarrow 3H_2O + 7914kJ$  is this endothermic or exothermic?

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 $2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$  energy endothermic the transfer of heat from the surroundings to the system energy  
 $2Hg(s) + 2Hg(l) \rightarrow O_2(g)$

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web  $q = 140\text{ g} \times 4.184\text{ J}/160\text{ C} \times q = 9372.16\text{ J}$   
 $35.5\text{ g} \times 1\text{ mole KCN} / 0.5453\text{ moles} = 65.1\text{ g}$   
 $0.5453\text{ mole moles} = 5$  when a 15 g sample of  $SnCl_2$  dissolves in 80 g of water in a calorimeter the temperature drops from 42 °C to 15 °C calculate the  $\Delta H$  for the process  
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web thermochemistry answers and explanations 1 a thermodynamic system has two means of exchanging energy with its surroundings heat flow and pressure volume work heat flow  $q$  is the transfer of thermal energy due to temperature difference work  $w$  is macroscopic energy transfer between the system and its surroundings that occurs

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web jan 10 2021  $\Delta H = 27.4\text{ kJ/mol}$  and  $\Delta S = 29.8\text{ J/K}$  sulfur trioxide  $SO_3$  is produced in large quantities in the industrial synthesis of sulfuric acid sulfur dioxide is converted to sulfur trioxide by reaction with oxygen gas write a balanced chemical equation for the reaction of  $SO_2$  with  $O_2(g)$  and determine its  $\Delta G$